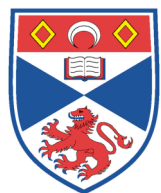


Solid-State NMR Studies of Molecular Motion

Sharon Ashbrook

School of Chemistry, University of St Andrews

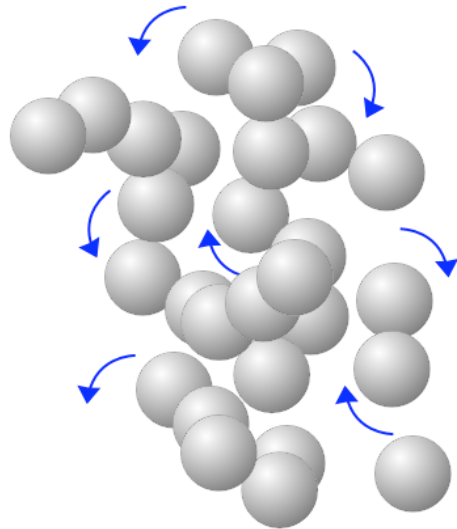


University
of
St Andrews

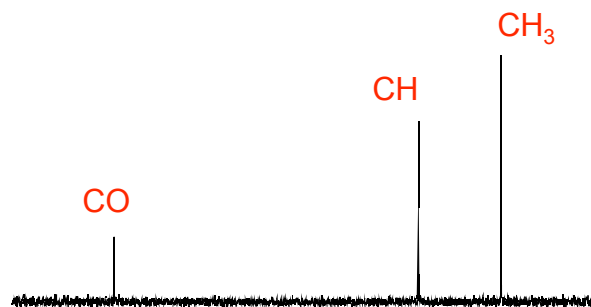
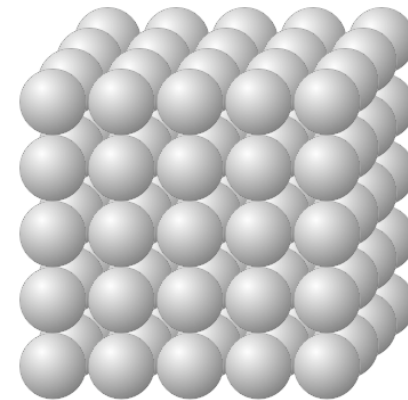


Motion in NMR

Solution



Solid



^{13}C NMR of alanine in solution

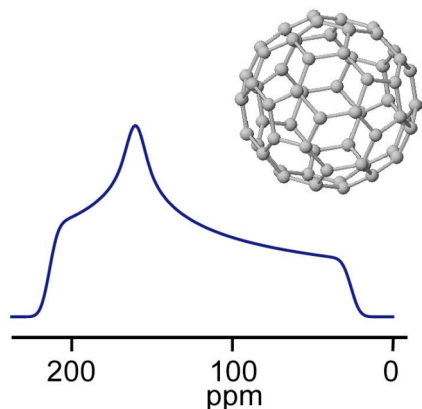


^{13}C NMR of solid alanine

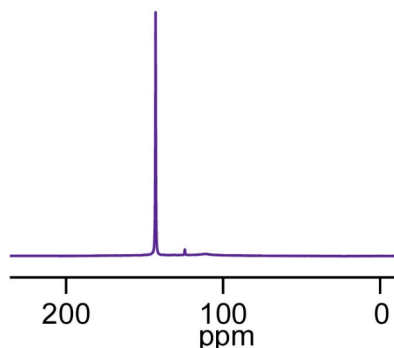
Motion in NMR

fullerene - carbon-13 NMR

low temperature

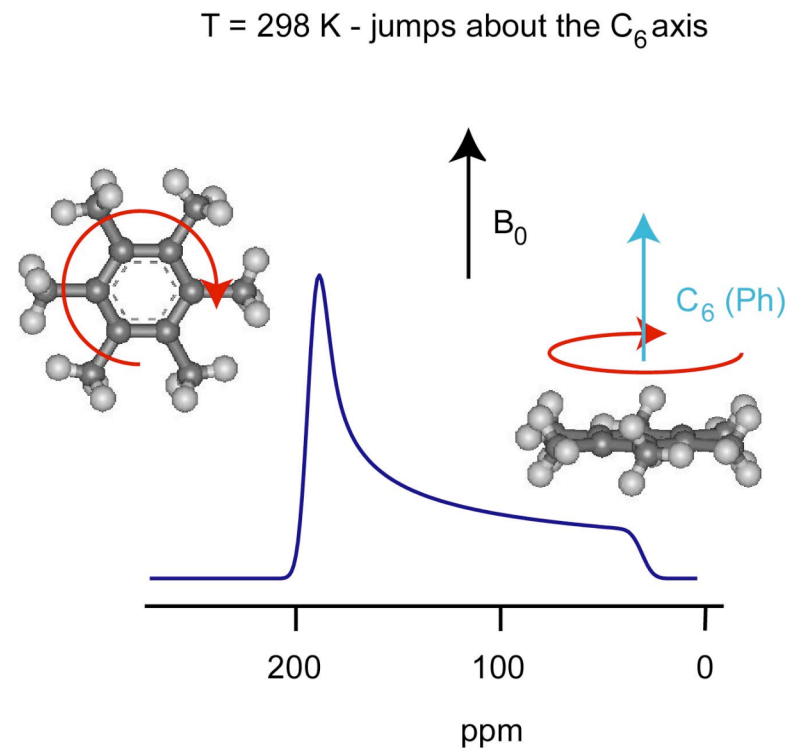
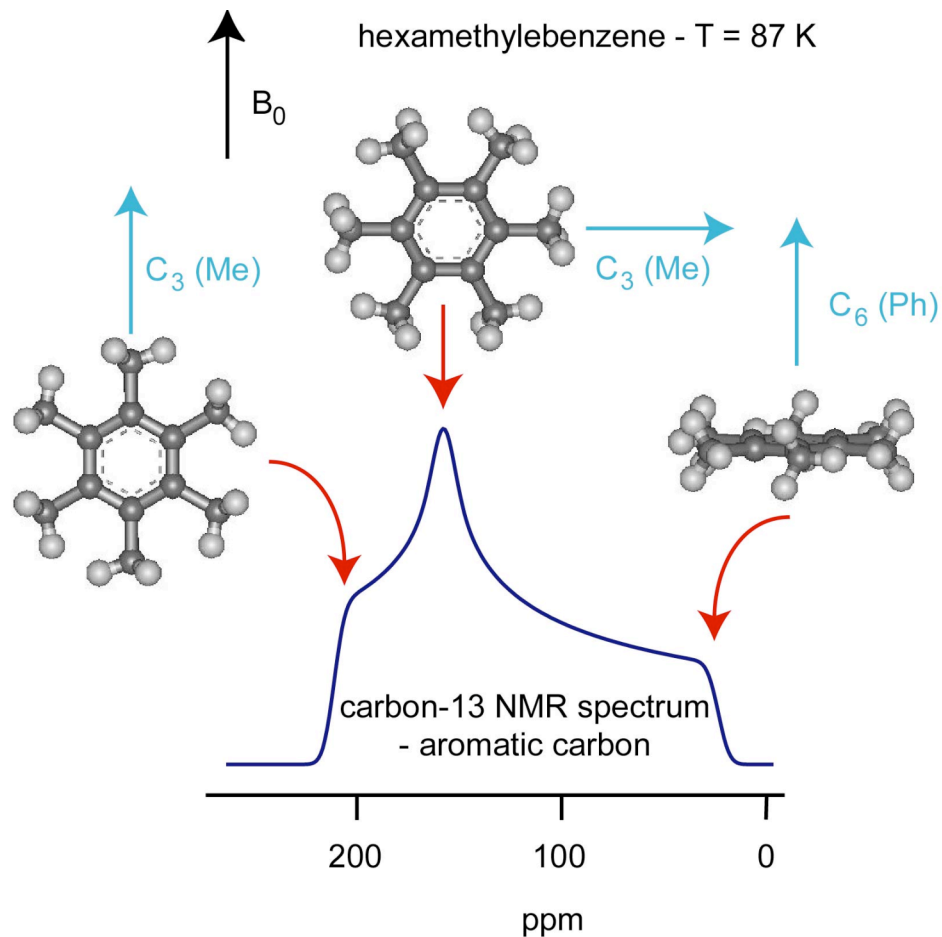


room temperature
tumbles isotropically



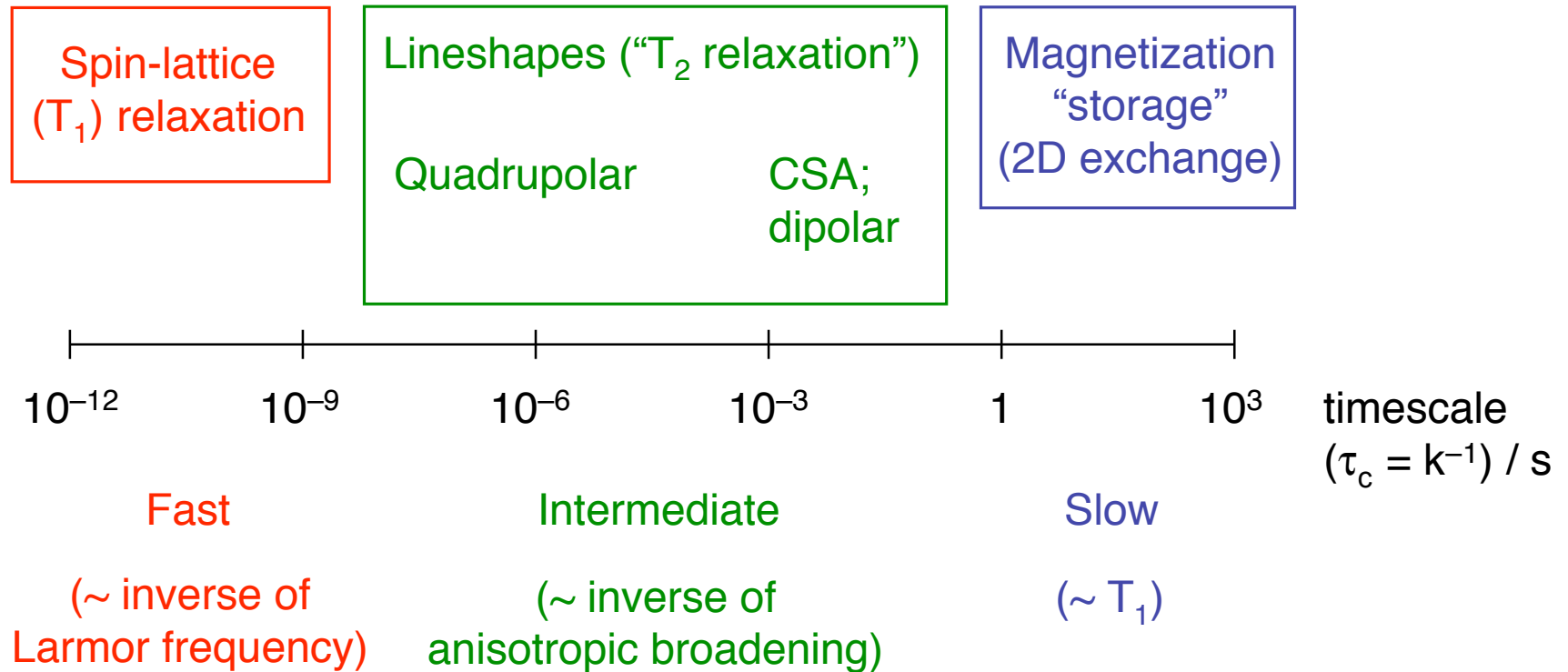
- C₆₀ is a “plastic crystal”
- At low temperature lineshape broadened by CSA
- At higher temperatures molecules tumble rapidly and isotropically on their lattice sites
- A single sharp, narrow line is observed

Motion in NMR

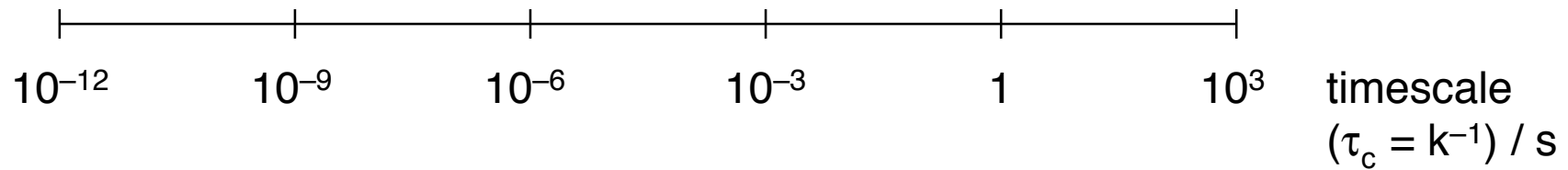


- At low temperature CSA tensor with three principal components
- At higher temperature C_6 jump averages two components of the tensor

Timescales



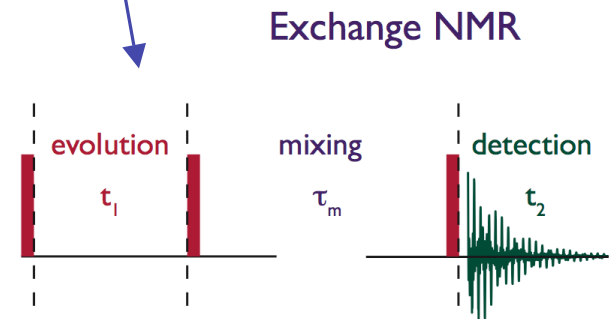
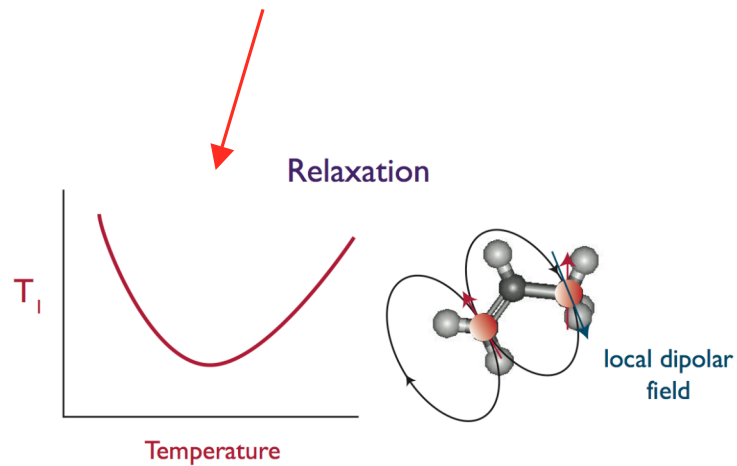
Timescales



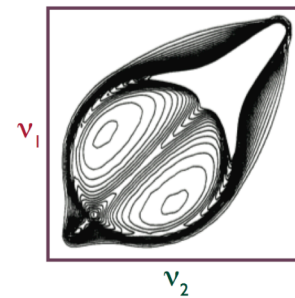
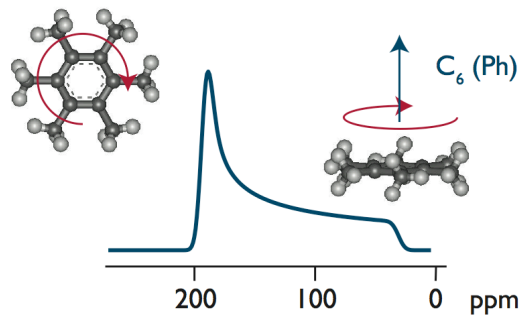
Fast

Intermediate

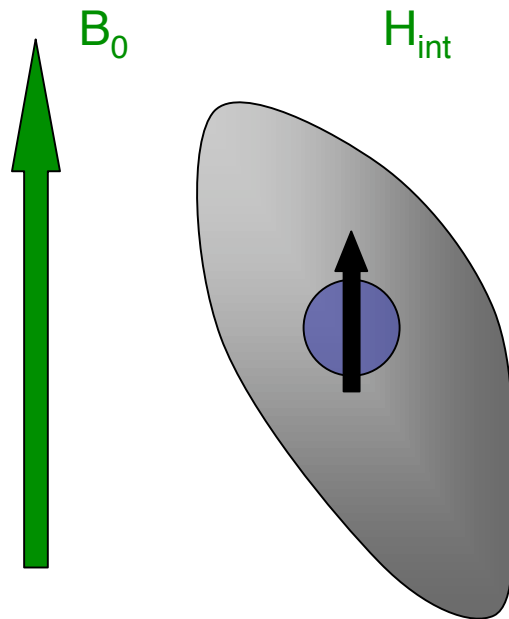
Slow



Lineshape Analysis

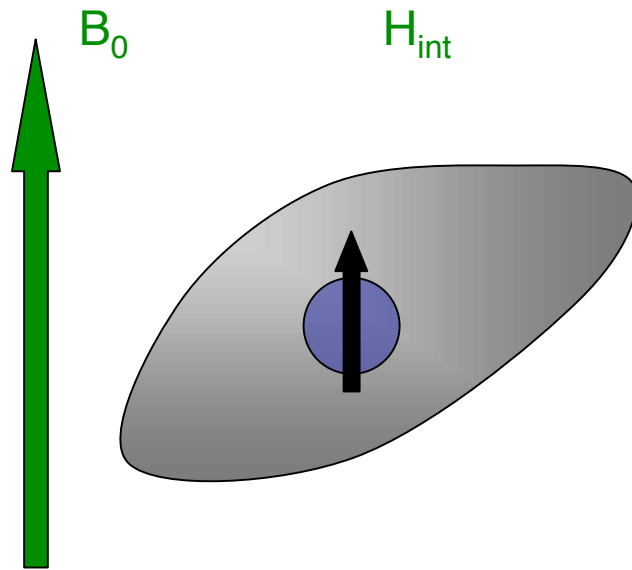


Rapidly fluctuating Hamiltonians



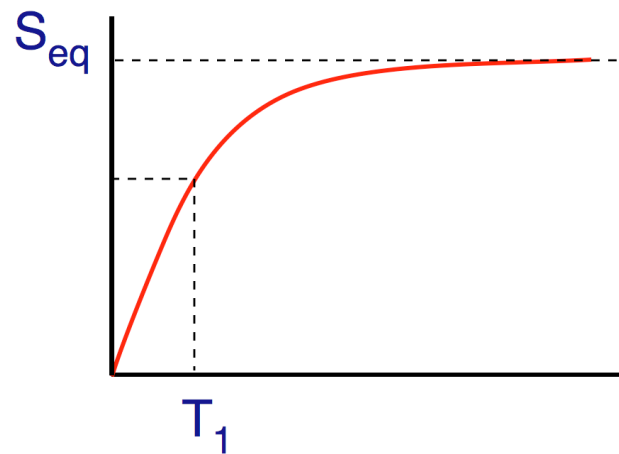
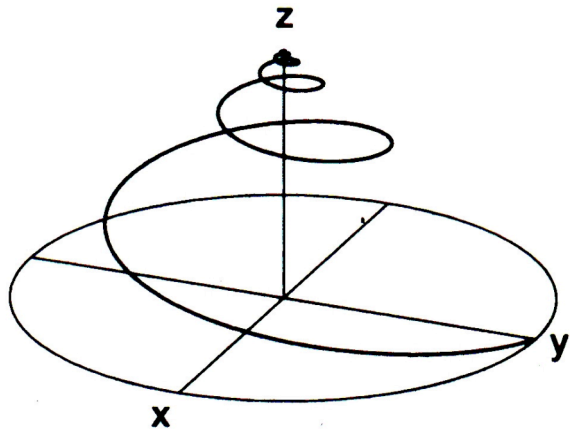
- Orientationally-dependent (anisotropic) interactions, H_{int}
- In liquids, rapid tumbling motion results in resonance frequency being determined only by isotropic or average value of H_{int}
- Also $\Delta m = \pm 1, \pm 2$ off-diagonal elements of H_{int} that cause $\Delta m = \pm 1, \pm 2$ transitions if motion is comparable with ω_0 or $2\omega_0$
- These random transitions give rise to the spin-lattice (T_1) relaxation and part of the spin-spin (T_2) relaxation behaviour

Rapidly fluctuating Hamiltonians

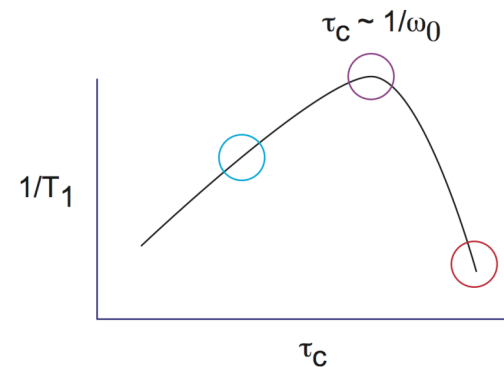
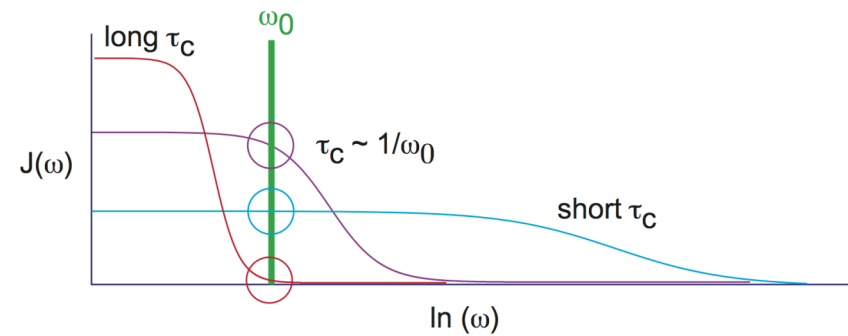
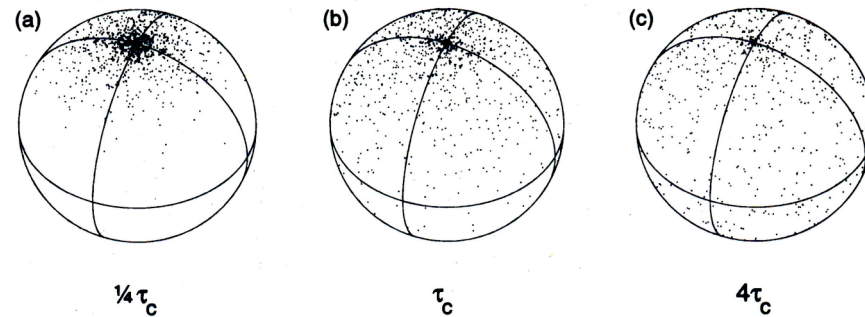


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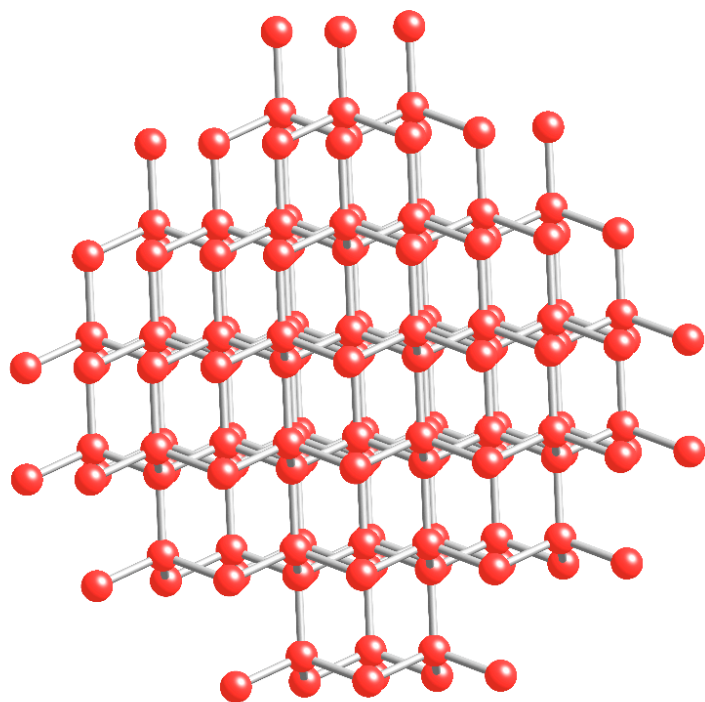
Spin-lattice (T_1) relaxation



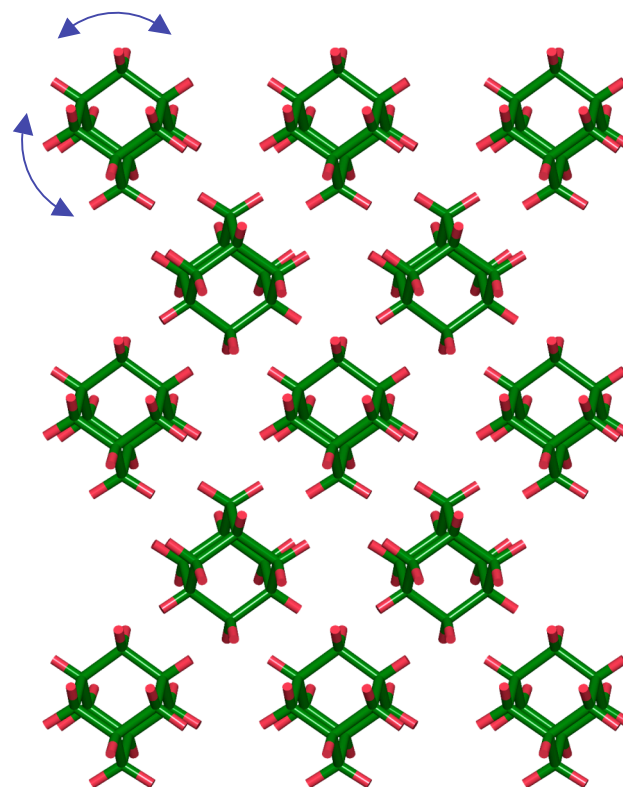
$$S(t) = S_{eq} (1 - \exp(-t/T_1))$$



Spin-lattice (T_1) relaxation

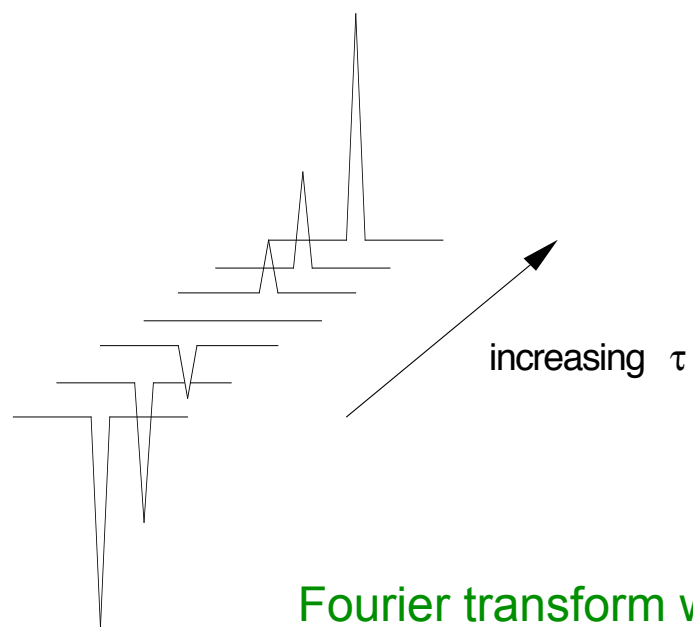
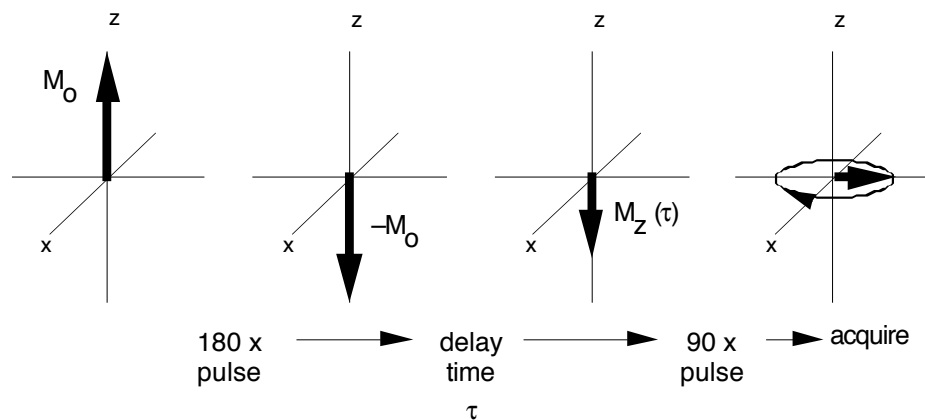


Diamond
 $T_1 \sim 24$ hours



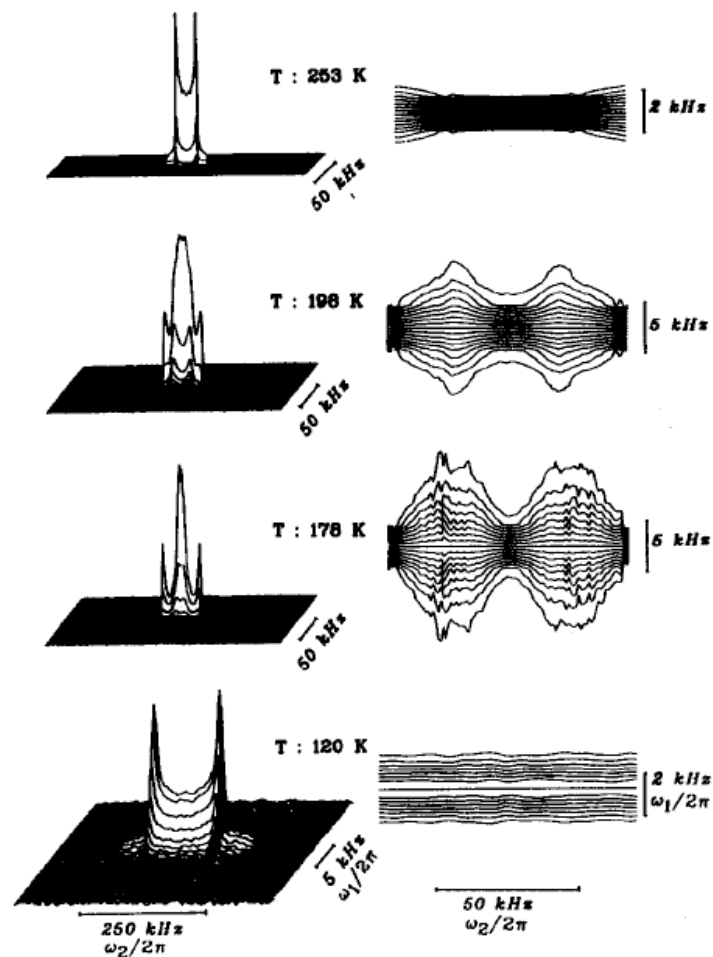
Adamantane
 $T_1 \sim 5$ s

Anisotropic T_1 in ^2H NMR



Fourier transform wrt τ

spinning methyl group
in d_3 -alanine



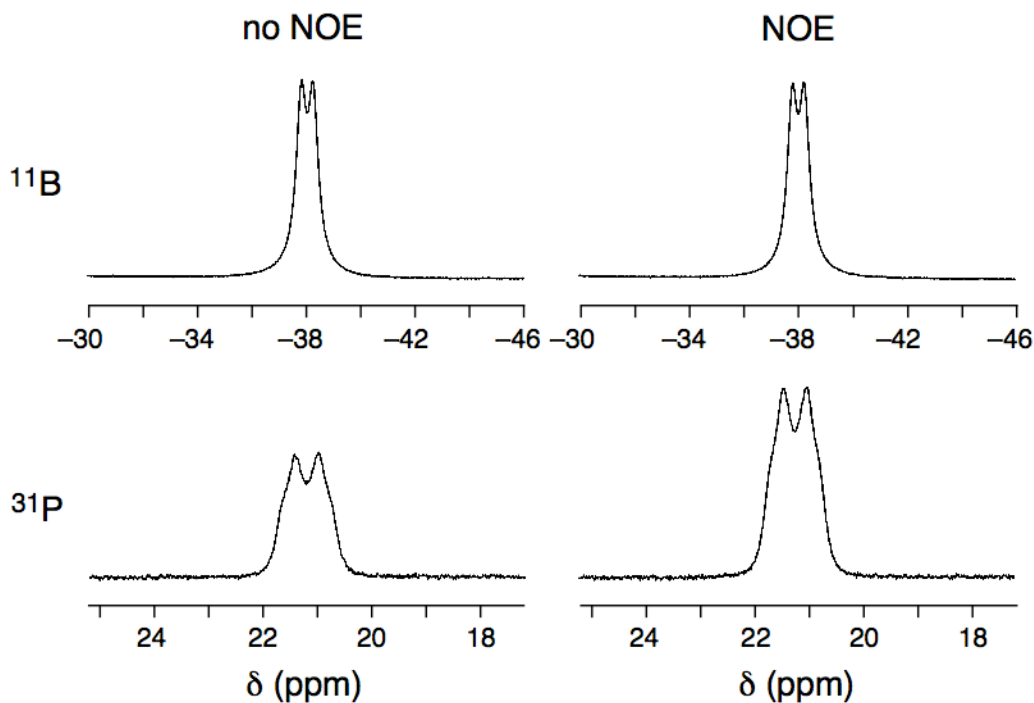
Schleicher et al., *J. Chem. Phys.* **92**, 6432 (1990)

NOE measurements

- NOE common in the liquid state for $I = 1/2$ nuclei
- Not usually observed for quadrupolar nuclei (efficient quadrupolar relaxation)
- Rare in solids as motion usually too slow

BH_3PPh_3

steady-state
NOE in CDCl_3



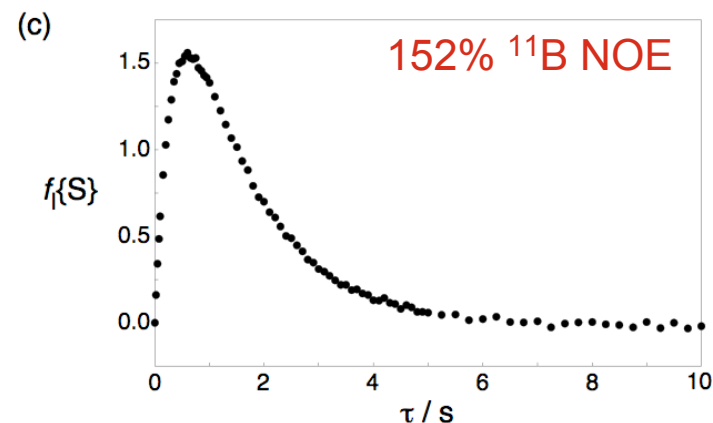
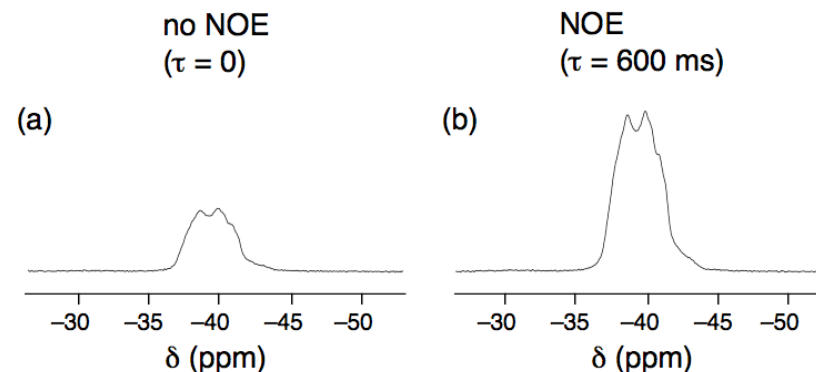
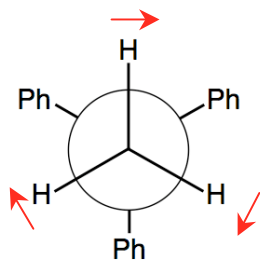
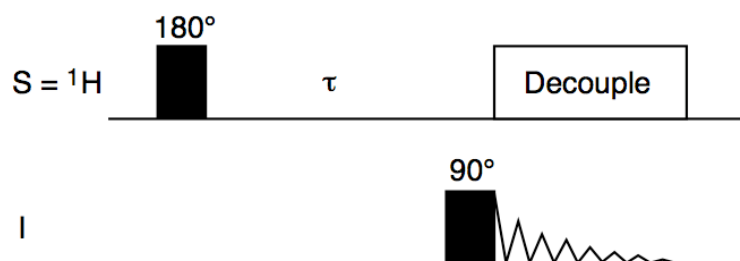
0% ^{11}B NOE

53% ^{31}P NOE

Ashbrook et al., *J. Am. Chem. Soc.* **128**, 6782 (2006)

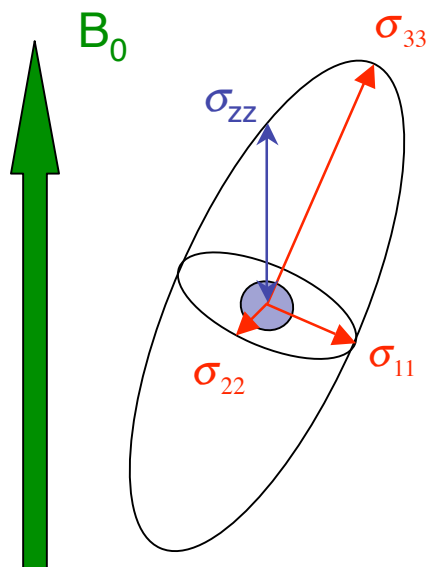
NOE measurements

transient NOE in BH_3PPh_3 solid



- Rapid BH_3 rotation modulates the dipolar interaction (to produce NOE) but not the quadrupolar interaction (so no rapid relaxation)

Slowly fluctuating Hamiltonians

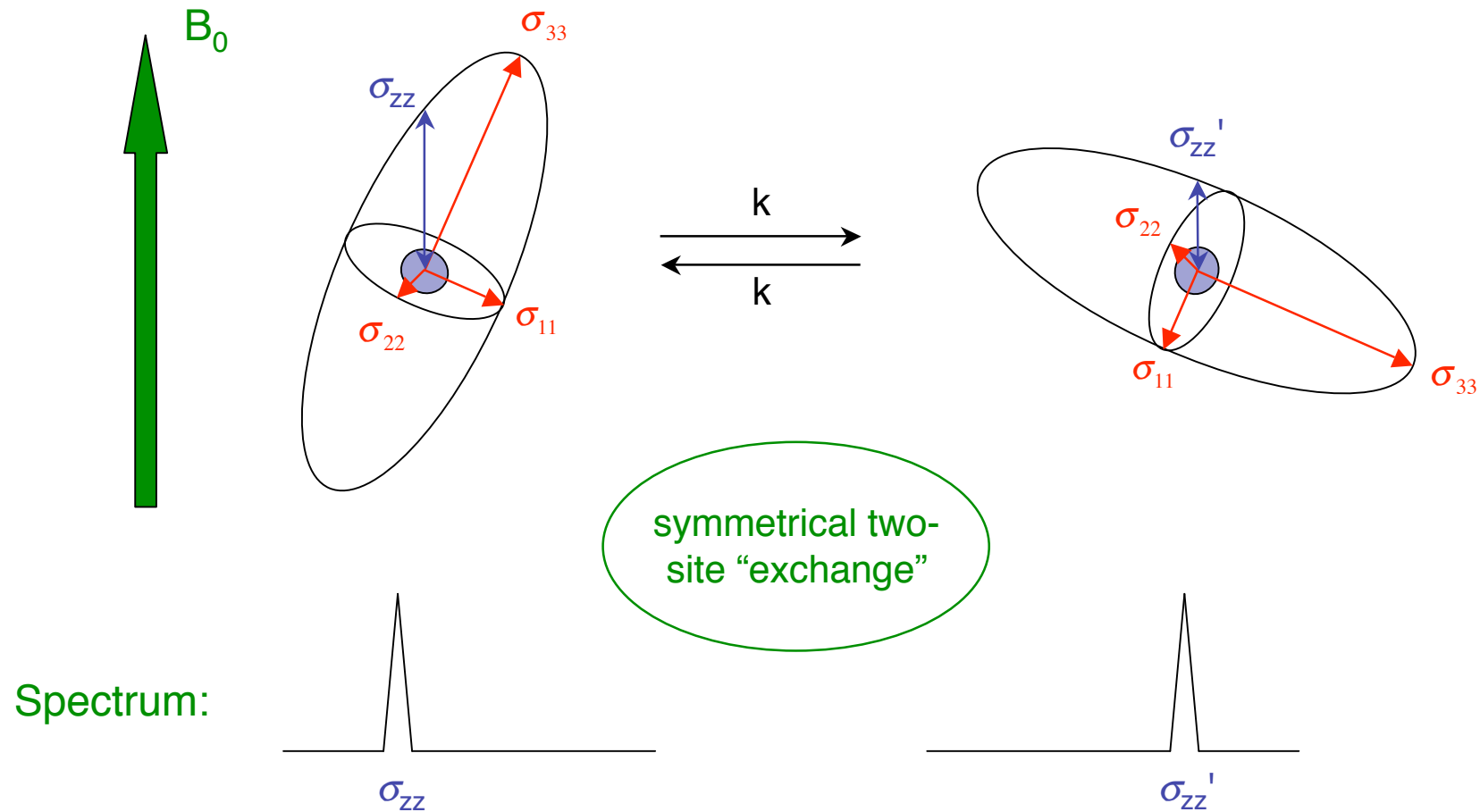


- To calculate resonance frequencies we truncate H_{int} (“make the secular approximation”, “retain only the terms that commute with H_Z ”)
- **Example:** the truncated shielding can be represented by an ellipsoid or shielding tensor, σ , that has three principal axes: σ_{11} , σ_{22} , and σ_{33}
- In a solid, observed shielding is given by σ_{zz}

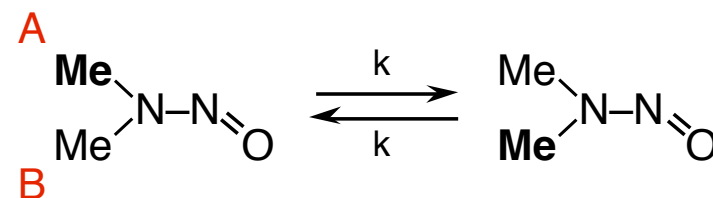
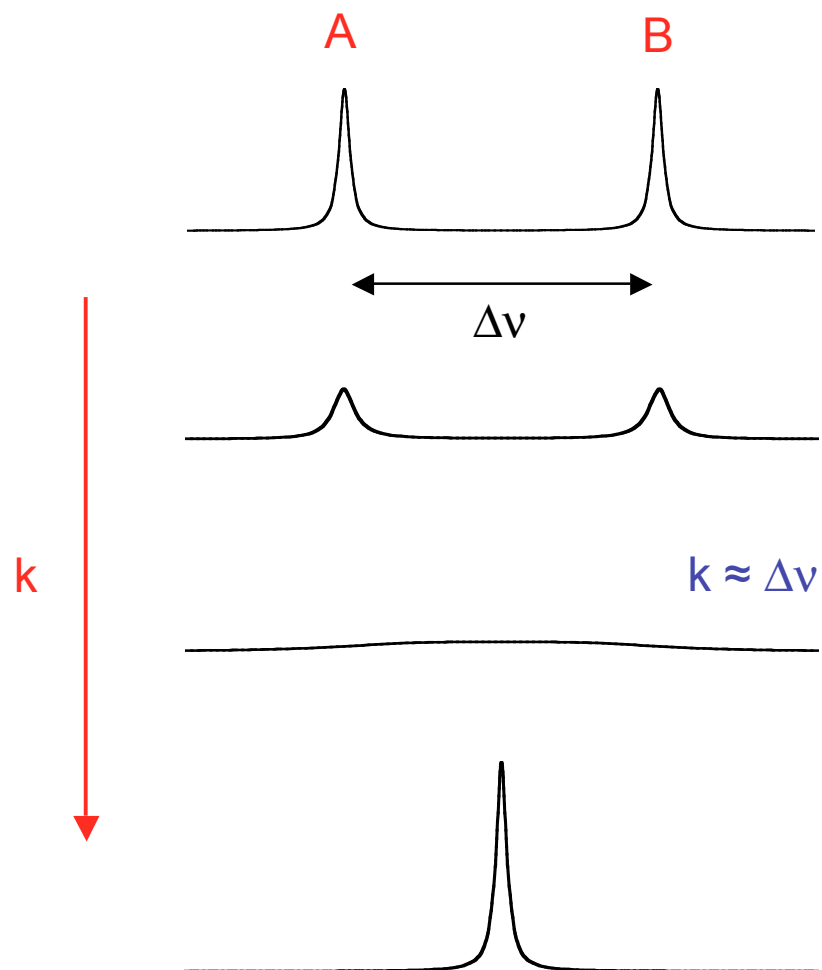
Isotropic shielding:

$$\sigma_{\text{iso}} = \sigma_{zz} = (1/3) (\sigma_{11} + \sigma_{22} + \sigma_{33})$$

Slowly fluctuating Hamiltonians

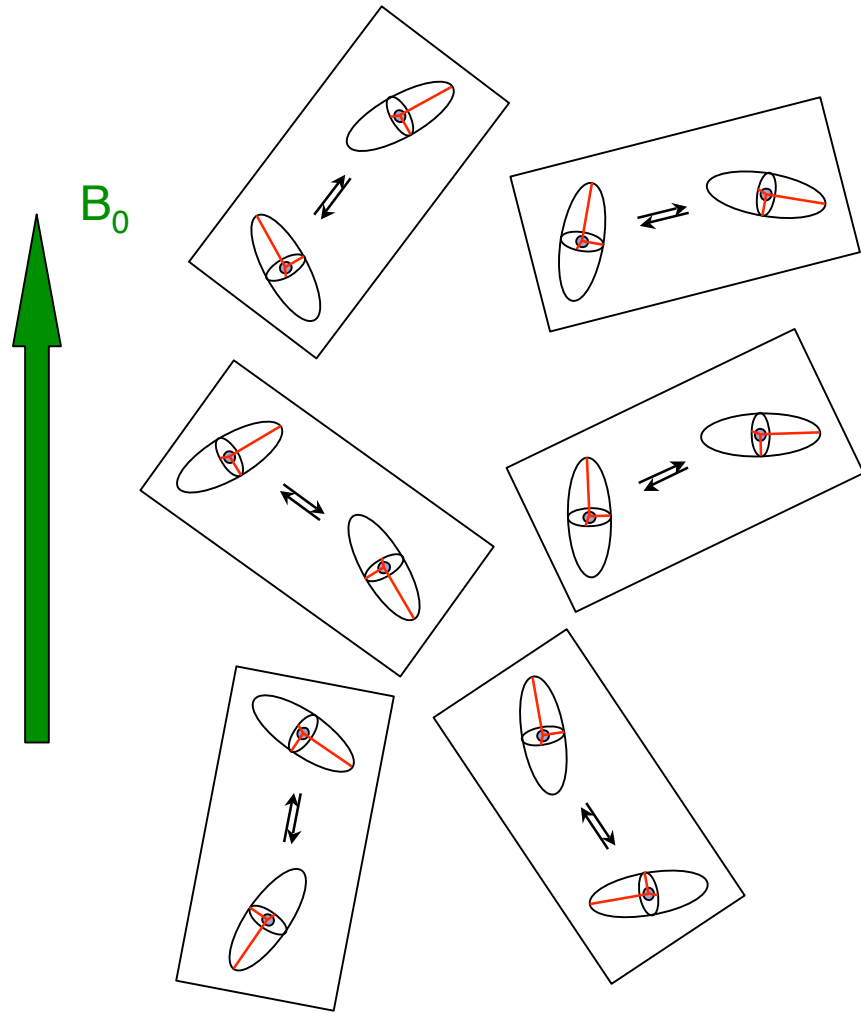


Exchange broadening



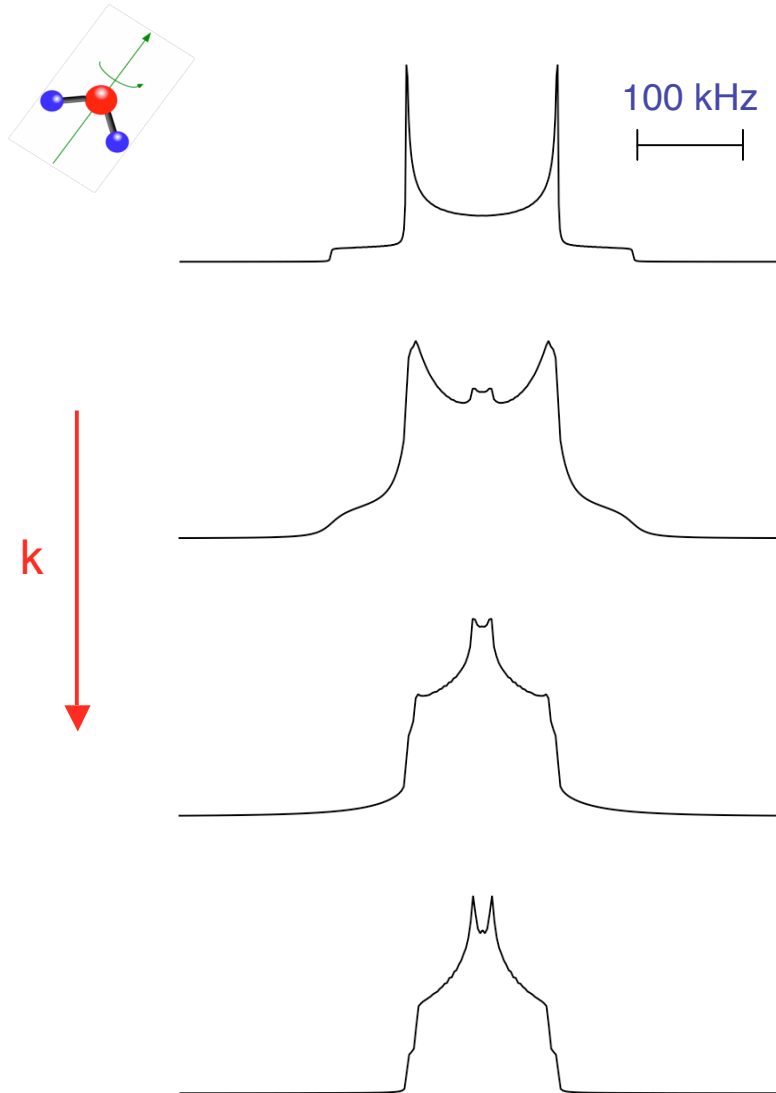
Dimethylnitrosamine –
the textbook example in
liquids NMR

Powder averaging and dynamics



- A powdered solid consists of microcrystals with a random, spherical distribution of orientations
- Therefore, exchange-broadened “pairs” must be averaged over all orientations to obtain powder spectrum

Powder averaging and dynamics

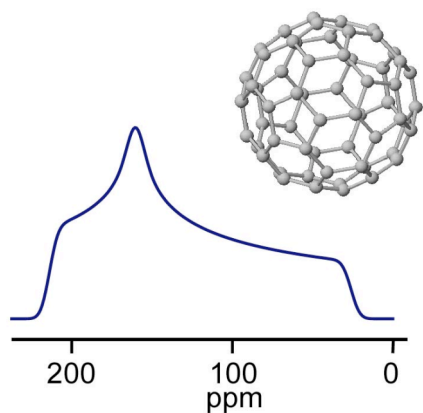


- **Example:** ^2H (spin $I = 1$) powder NMR spectrum as function of k
- Changes in powder lineshape and underlying linewidth (“ T_2 relaxation”) are best observed using echo methods (e.g., quadrupolar echo)
- Very powerful method of testing mechanism of reorientation and determining k
- “Exchange broadening” and “ T_2 relaxation caused by slow motions” can be thought of as the same thing

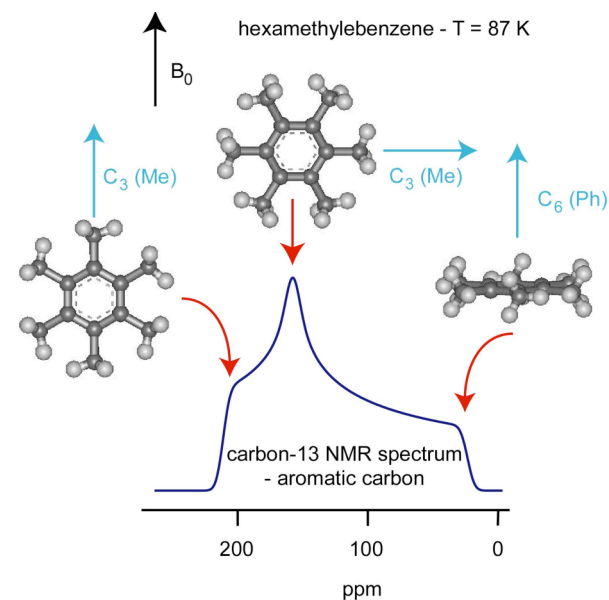
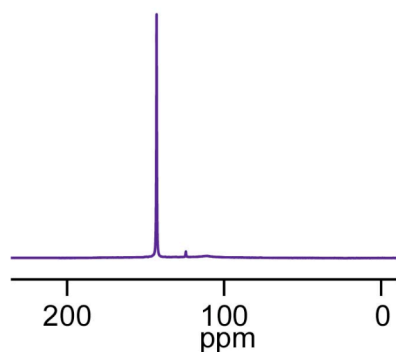
Powder averaging and dynamics

fullerene - carbon-13 NMR

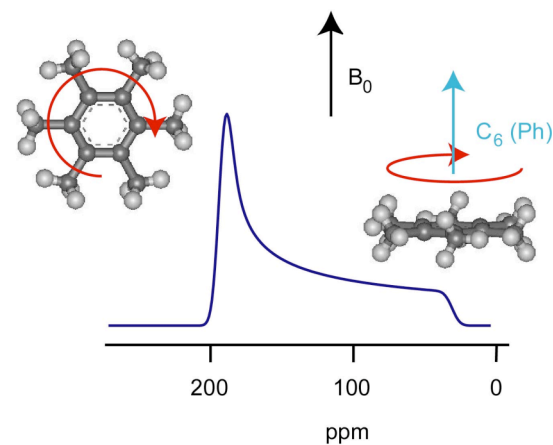
low temperature



room temperature
tumbles isotropically



T = 298 K - jumps about the C₆ axis



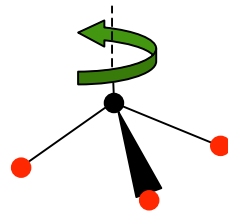
Powder averaging and dynamics



109° 2-site

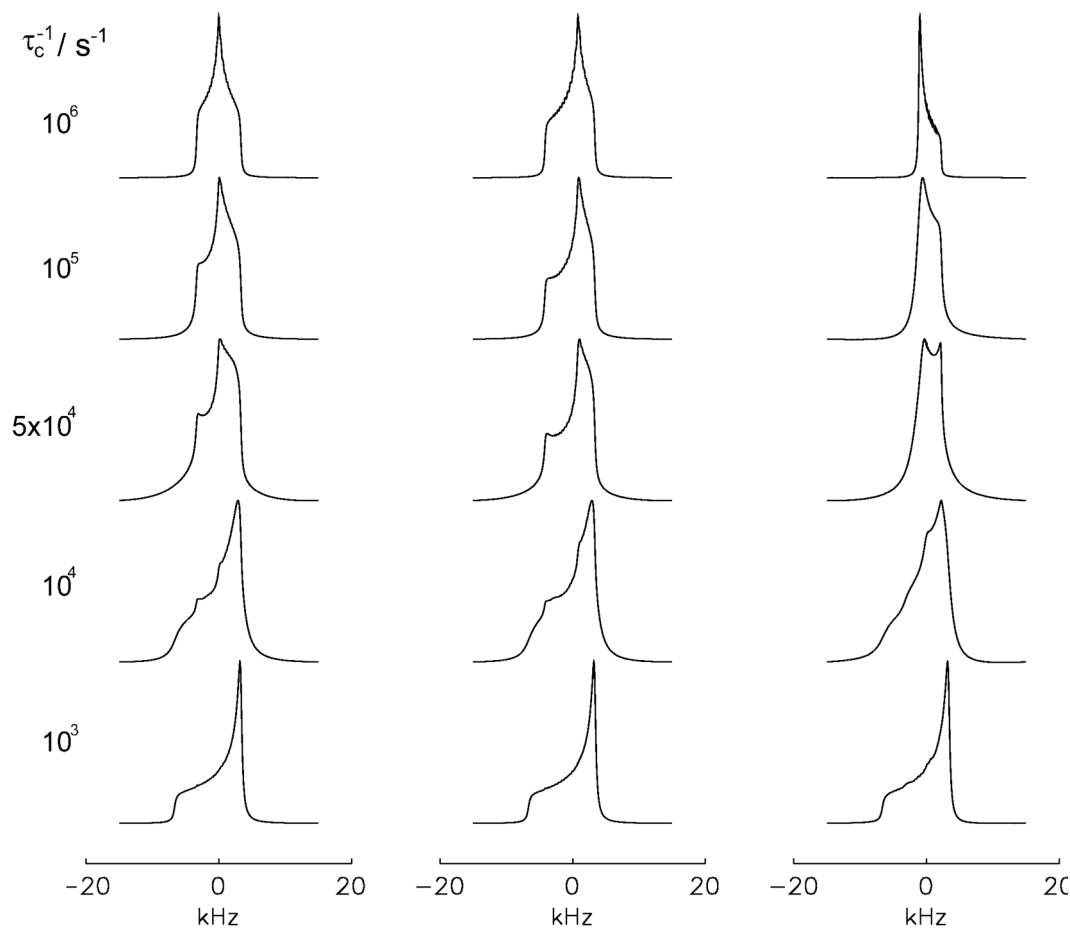


120° 2-site

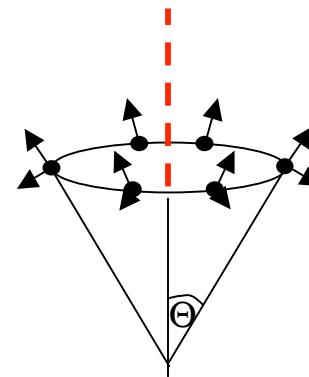


109° 3-site

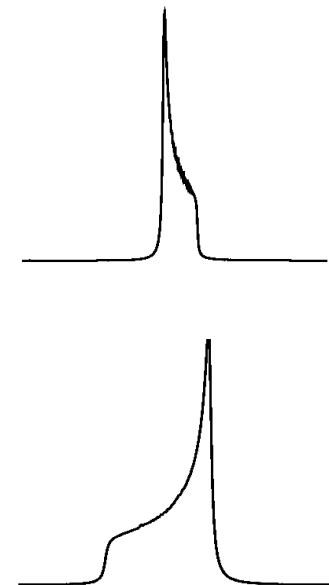
Effective CSA interaction tensor is the **AVERAGE** over all the sites visited during the motion



Axial motion:



$\eta = 0$
axially symmetric



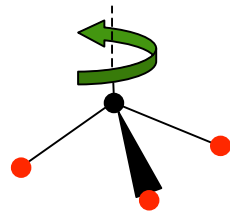
Powder averaging and dynamics



109° 2-site

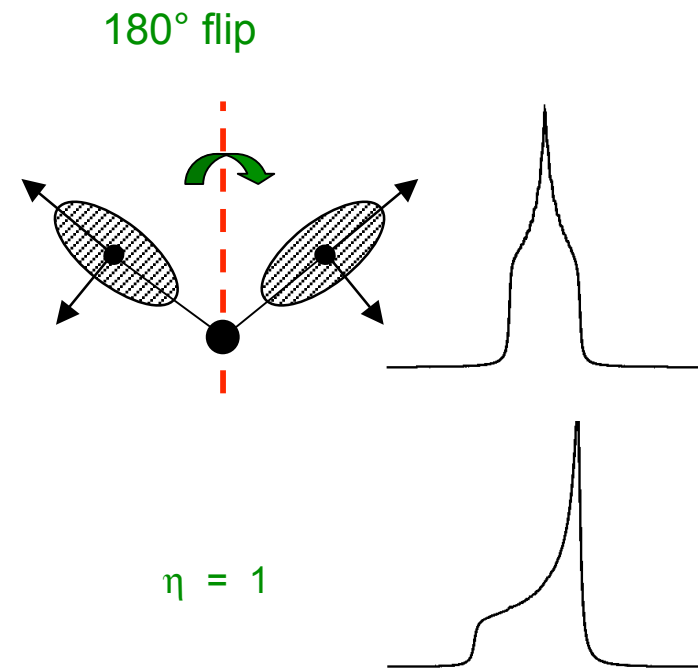
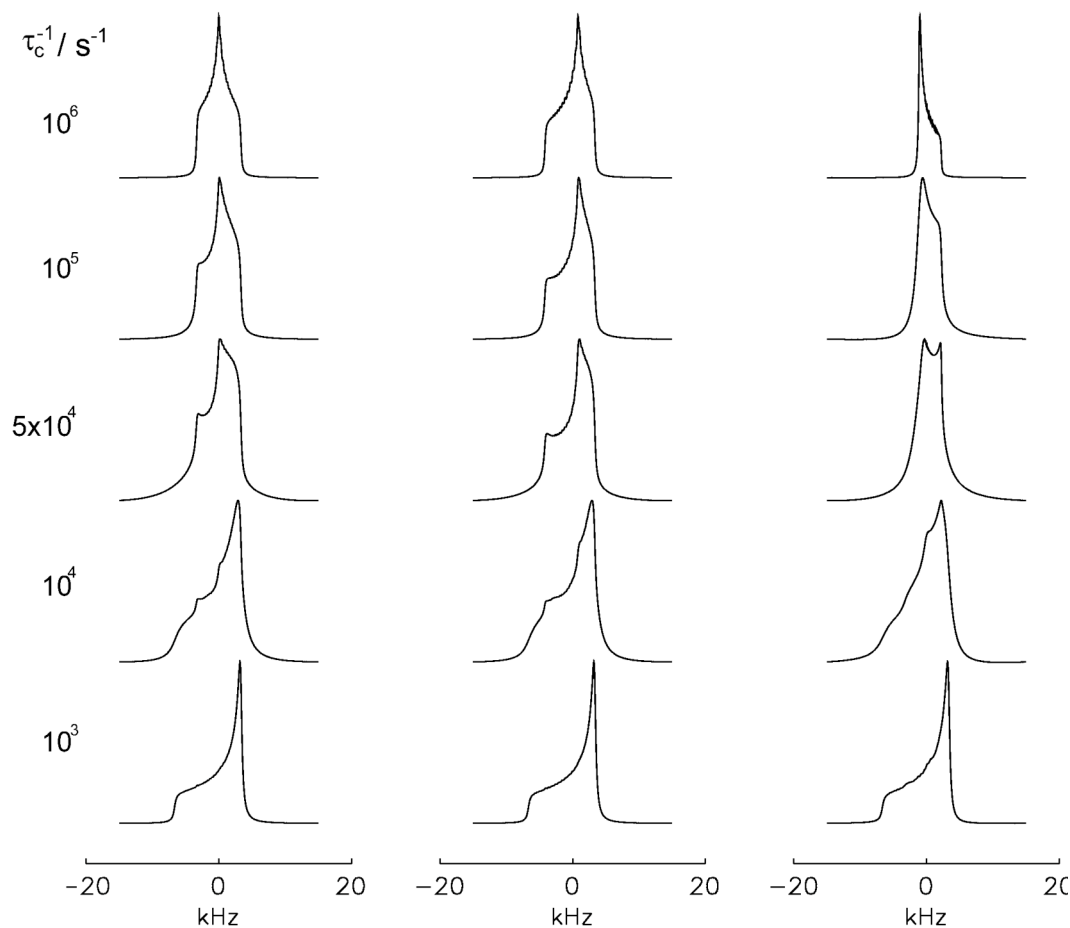


120° 2-site



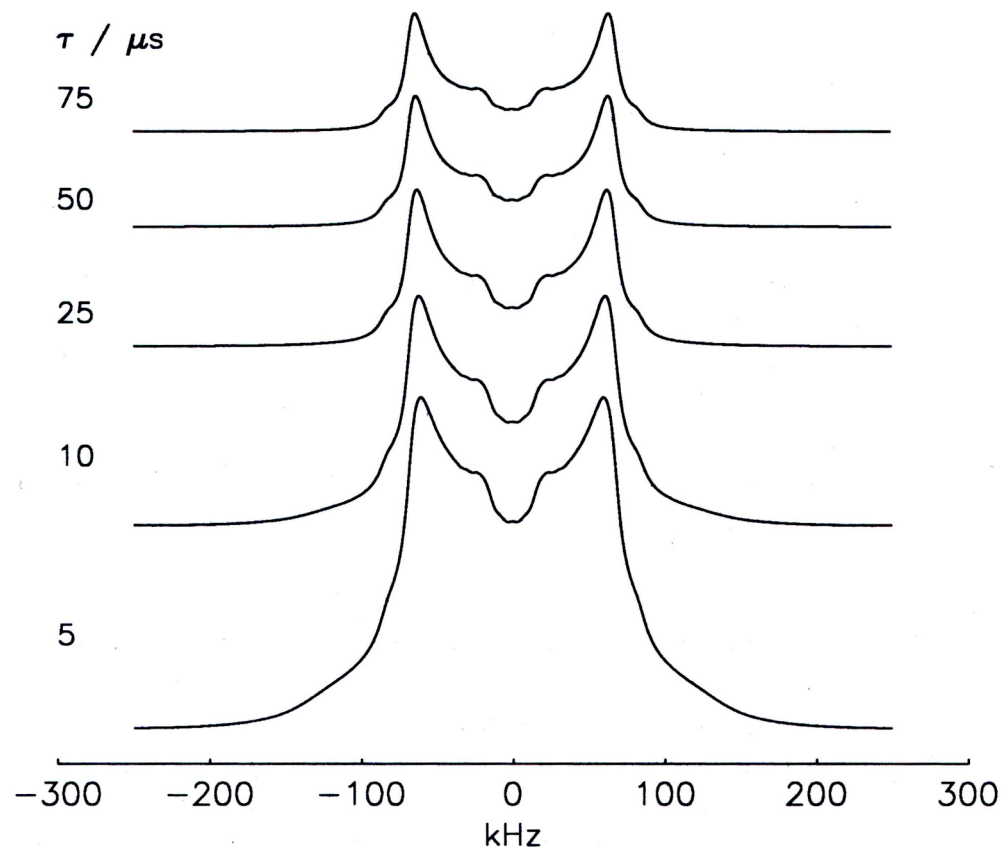
109° 3-site

Effective CSA interaction tensor is the **AVERAGE** over all the sites visited during the motion



NB- Change to handout

Powder averaging and dynamics

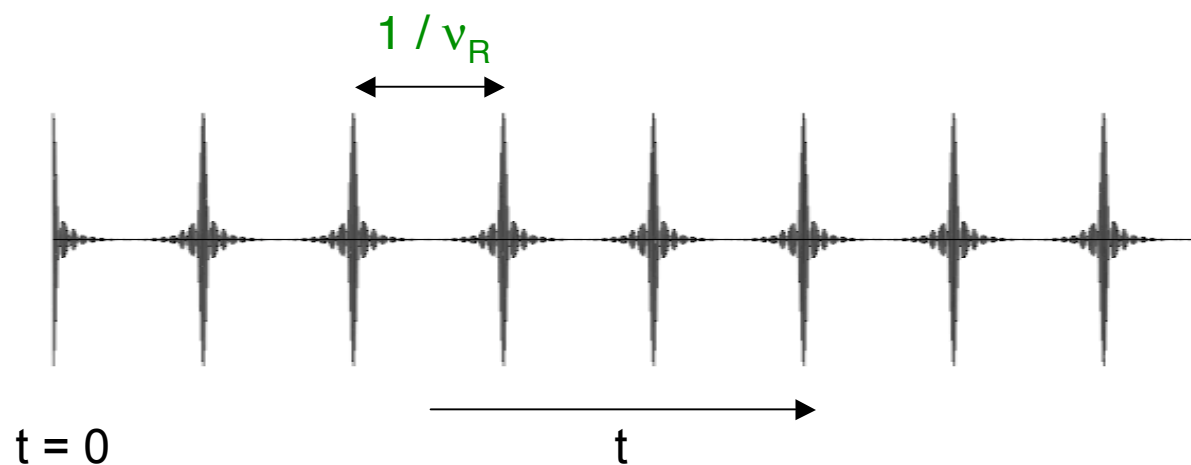


- In addition to lineshape changes with temperature motion can also be probed by lineshape changes with echo delay
- Anisotropic “ T_2 ” relaxation depending upon type of motion and orientation with respect to B_0

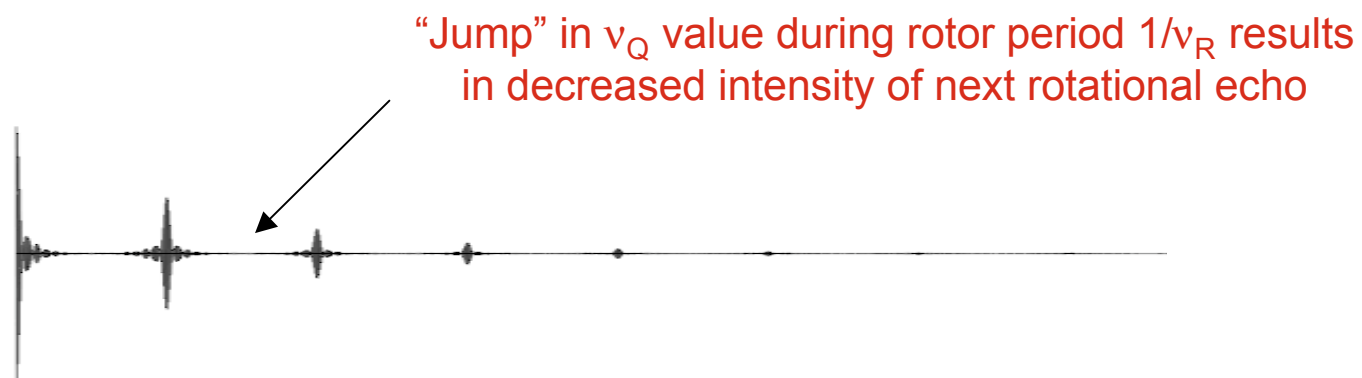
^2H two-site exchange with
hopping angle = 120° , with
 $\tau_c = 10^{-5} \text{ s}$

Dynamics and MAS

no motion

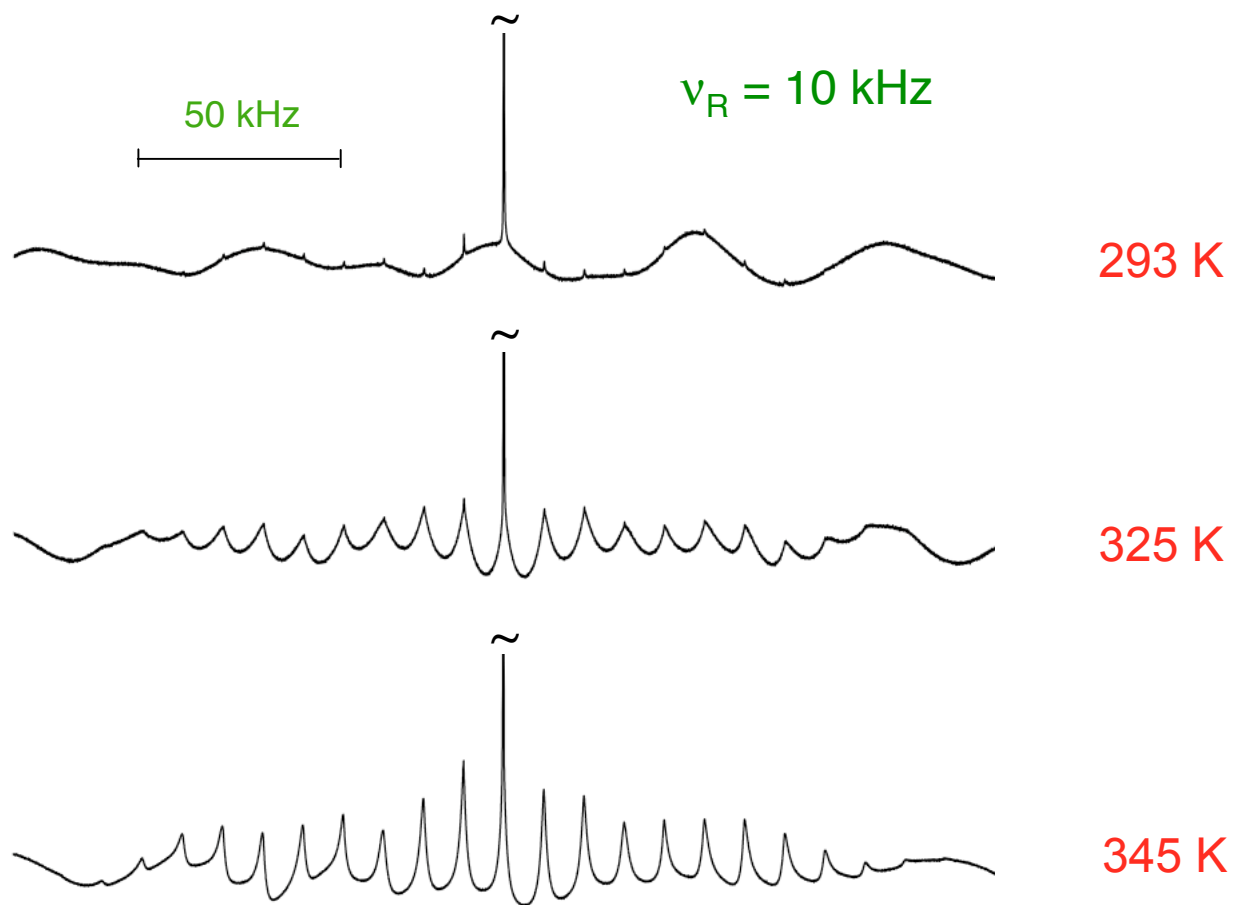


motion

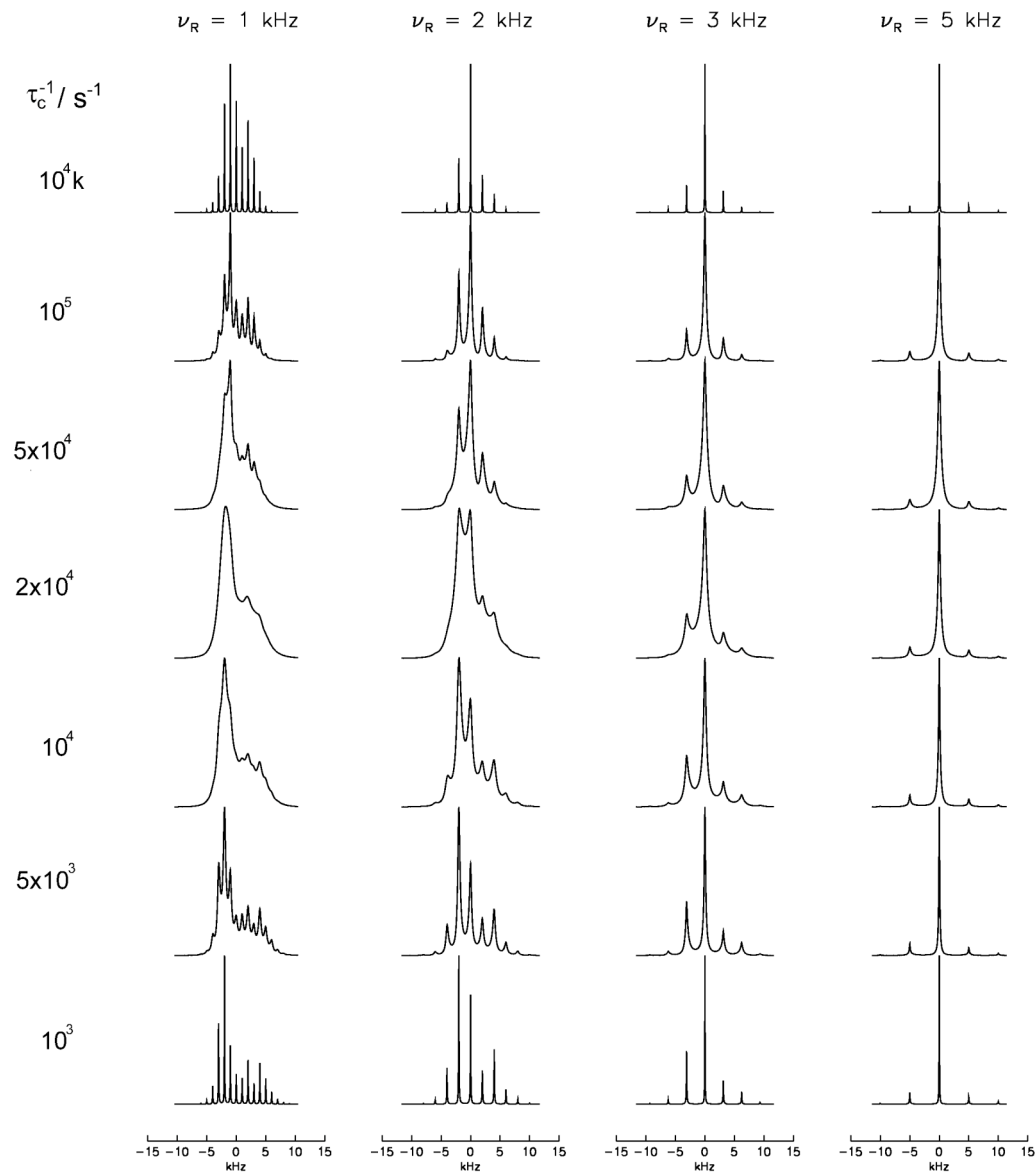


Dynamics and MAS

^2H MAS of $\text{Na}_2\text{S}_4\text{O}_6 \cdot 2\text{D}_2\text{O}$

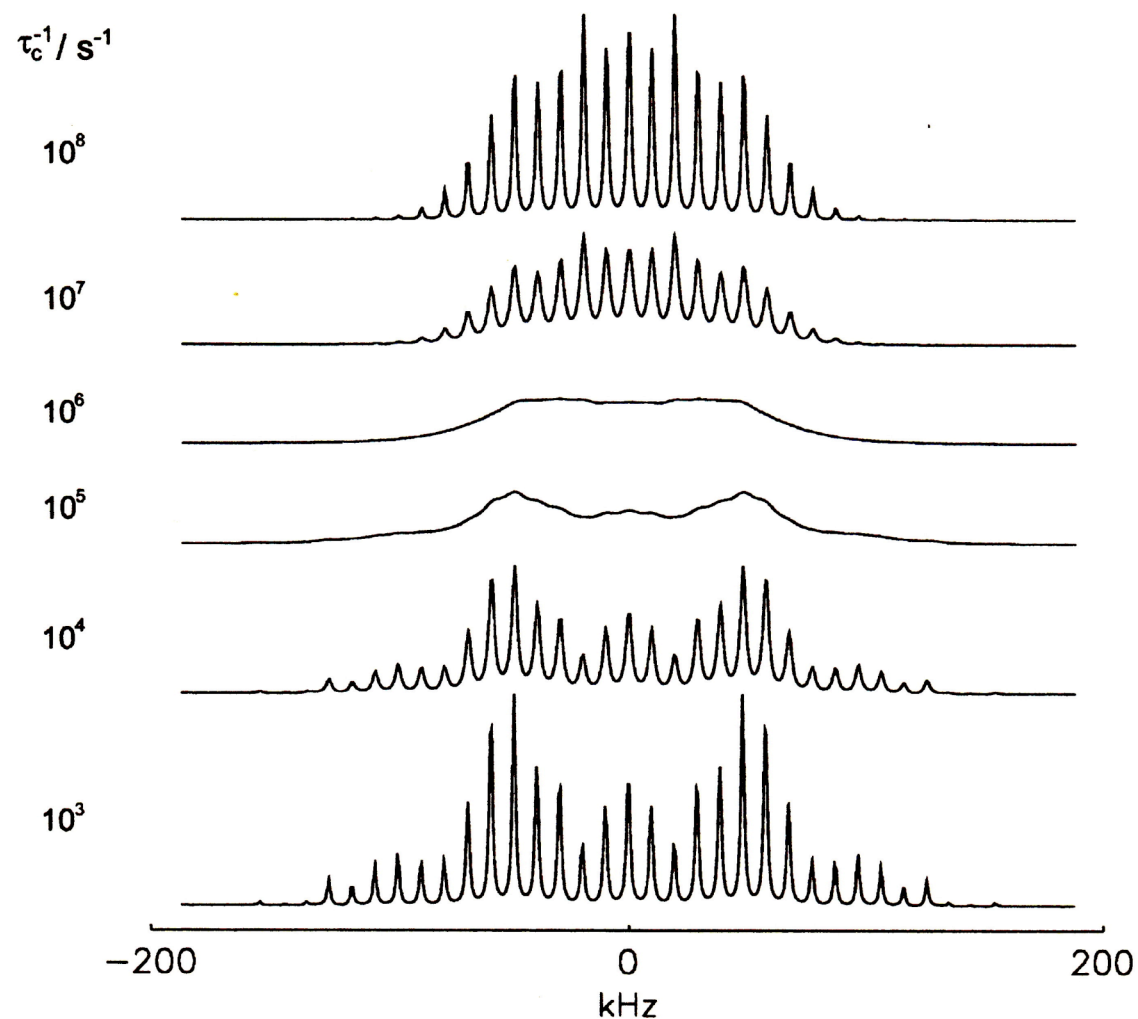


Dynamics and MAS



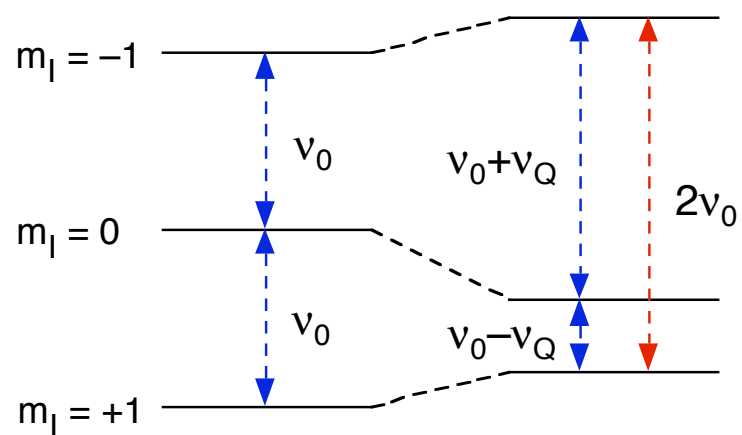
Typical CSA lineshapes
under two-site exchange
with hopping angle of 120°

Dynamics and MAS

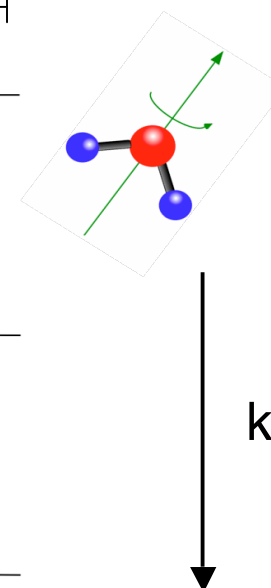
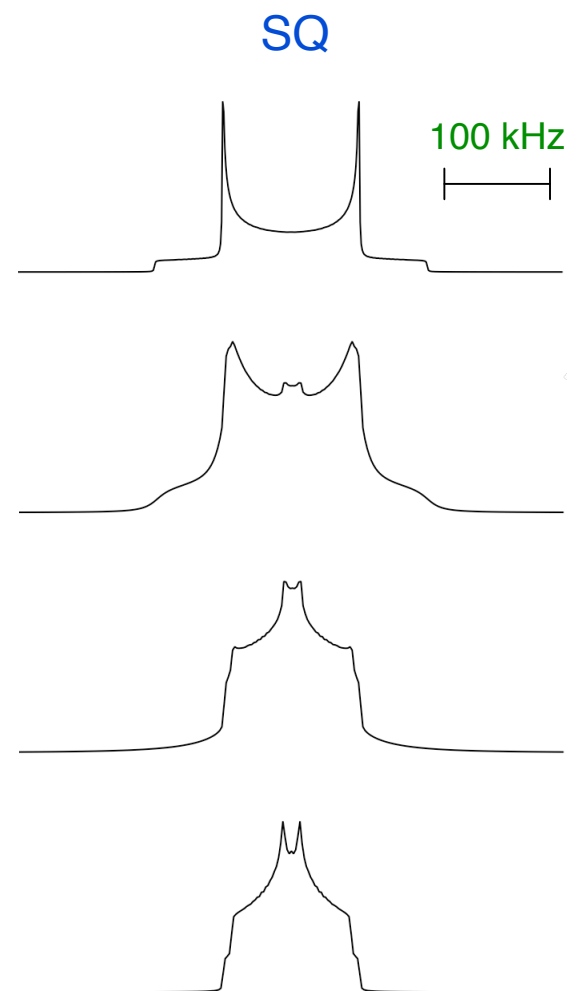
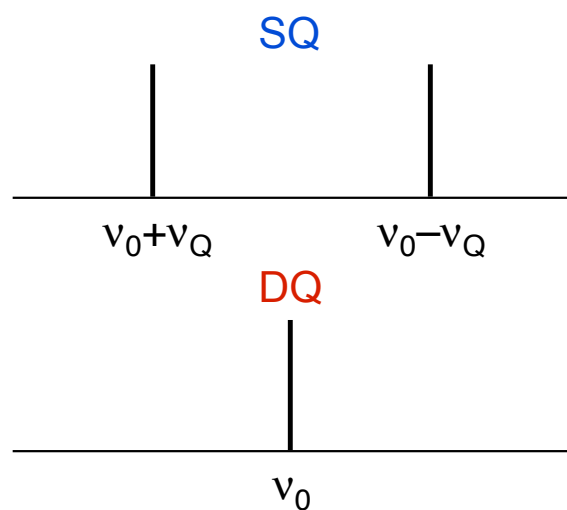


^2H lineshapes under two-site exchange with hopping angle of 120°

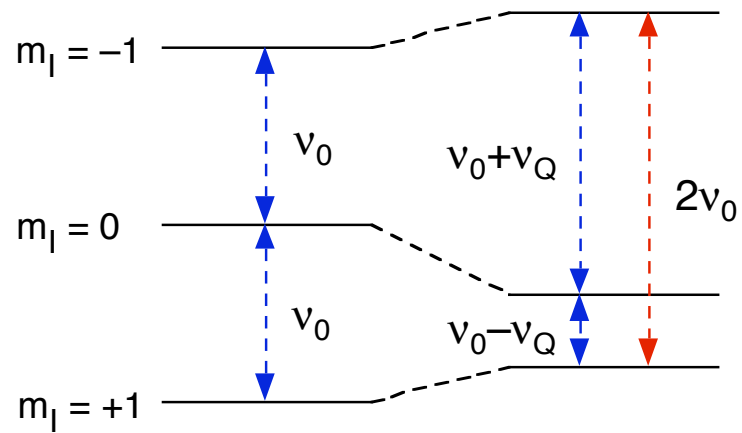
Double-quantum MAS



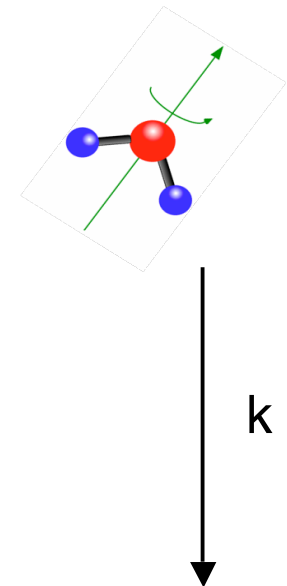
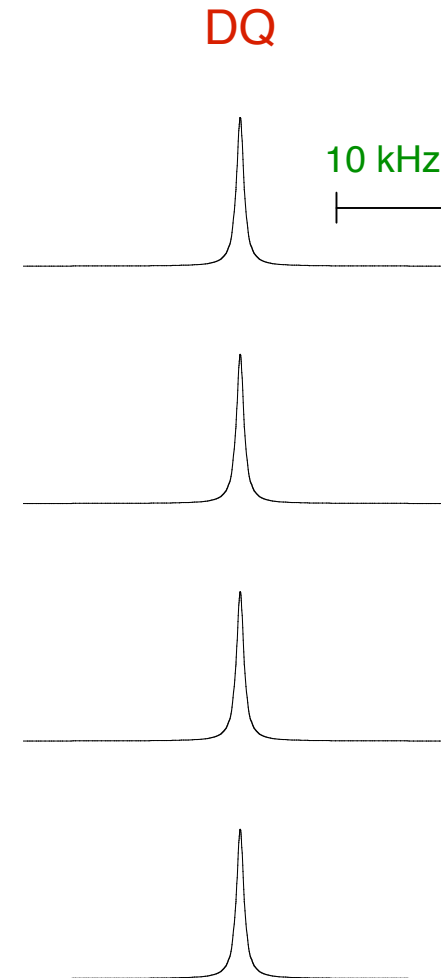
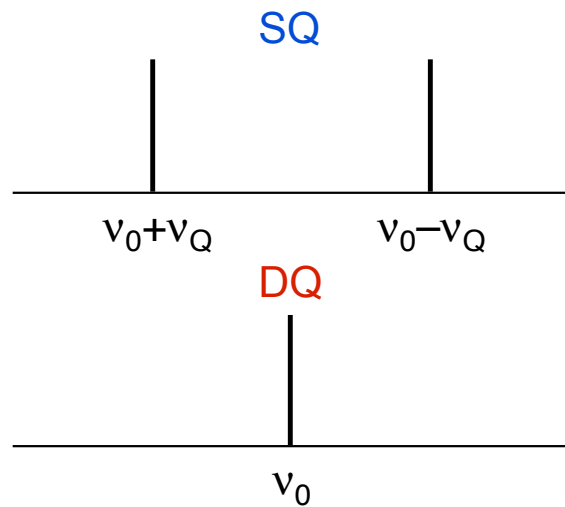
Zeeman + Quadrupolar



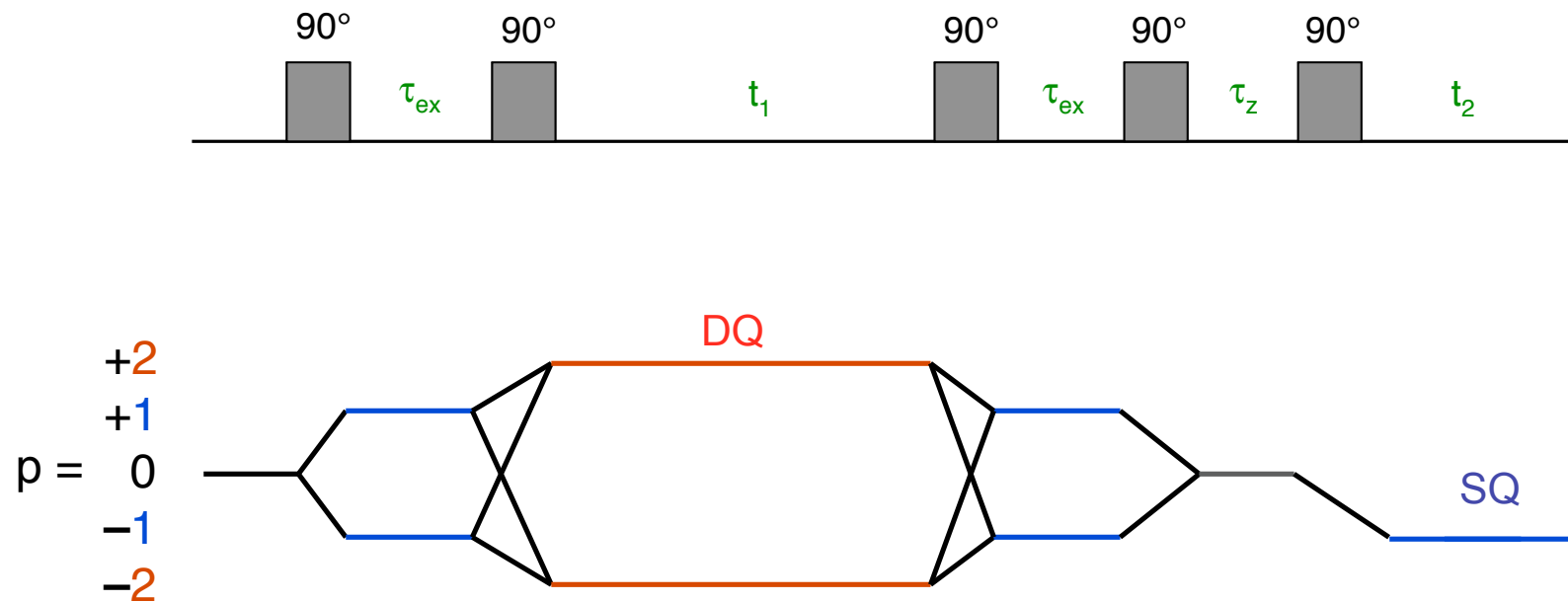
Double-quantum MAS



Zeeman + Quadrupolar

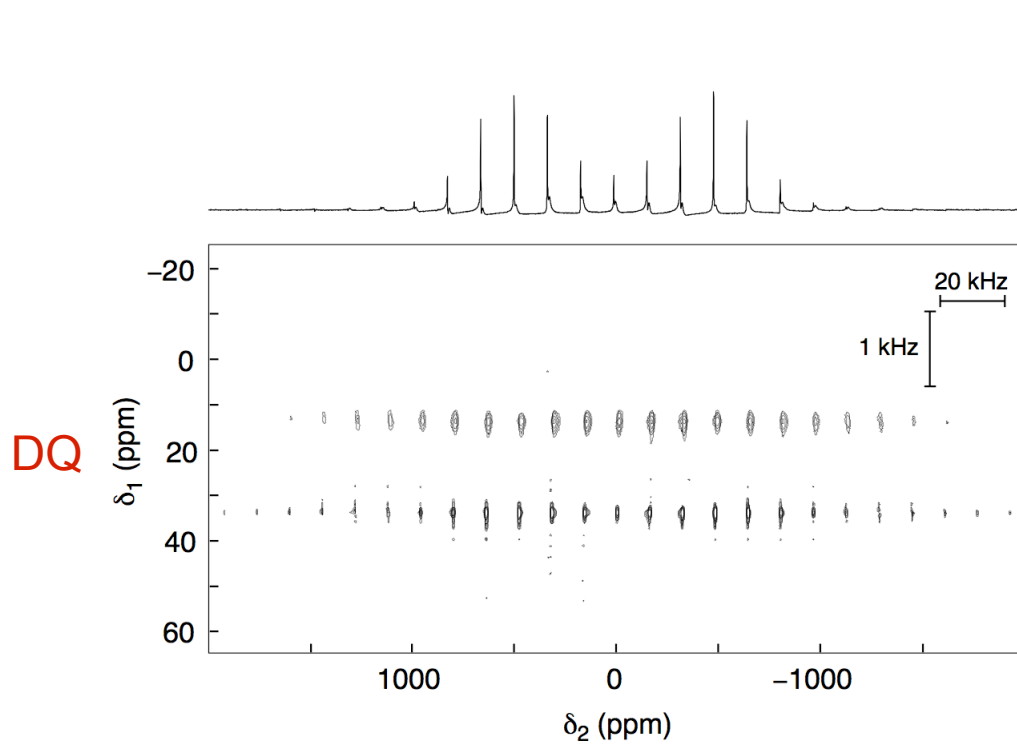


Double-quantum MAS



Double-quantum MAS

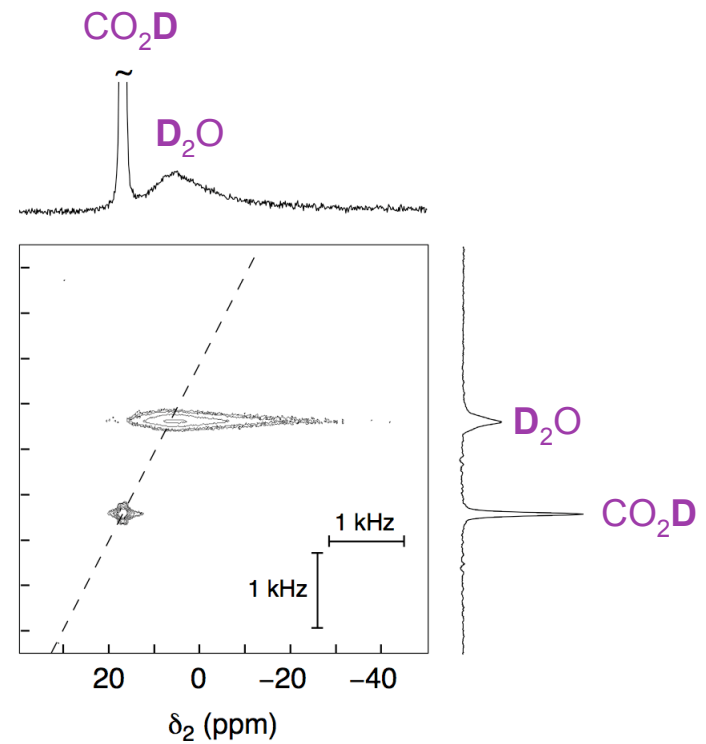
^2H DQMAS oxalic acid dihydrate



DQ

SQ

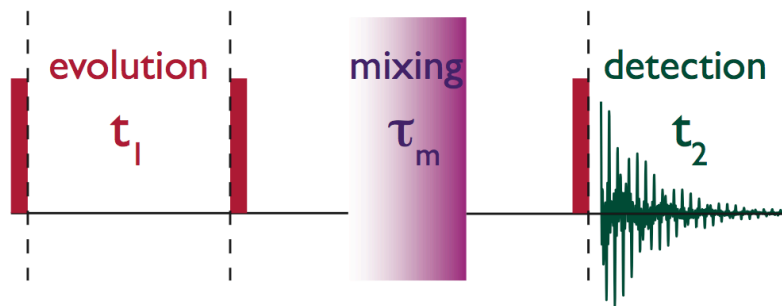
not rotor
synchronized in t_2



SQ

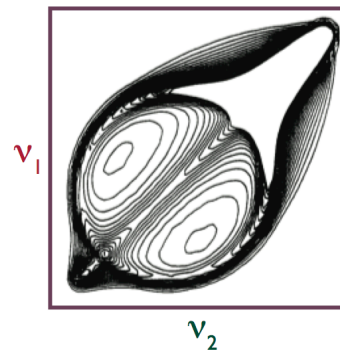
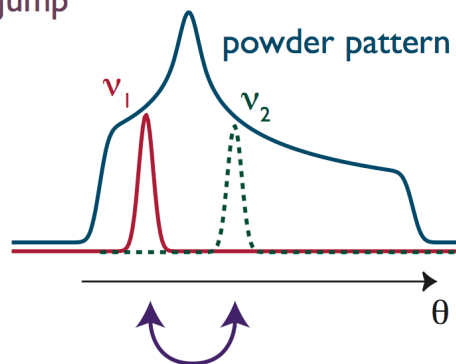
rotor
synchronized in t_2

Exchange experiments



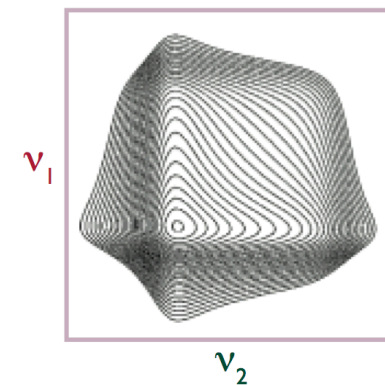
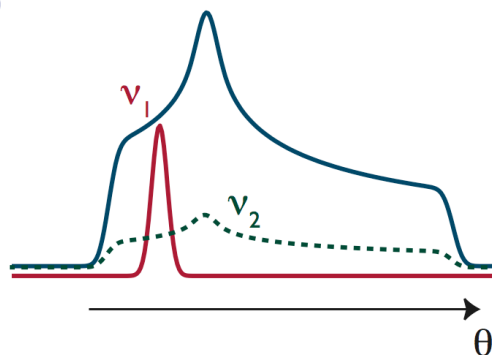
Typical two-dimensional exchange experiment

well-defined jump

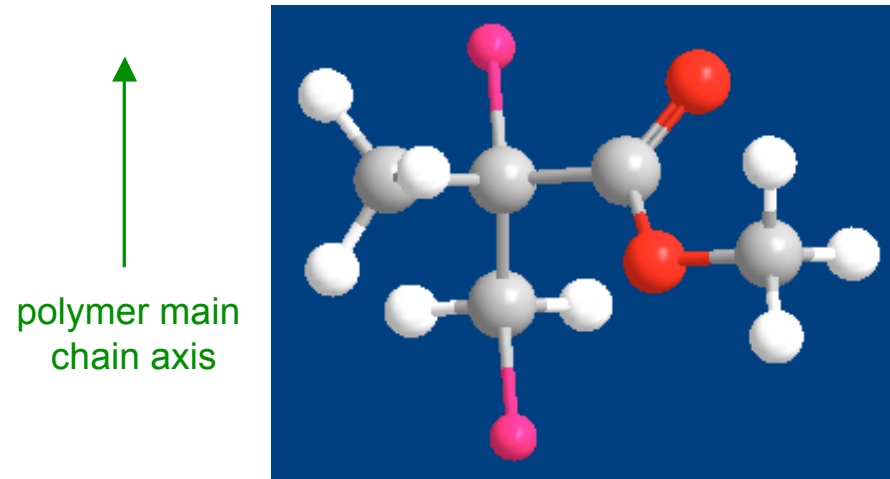


For a single site, static experiments can provide information on the type of reorientation

random jump

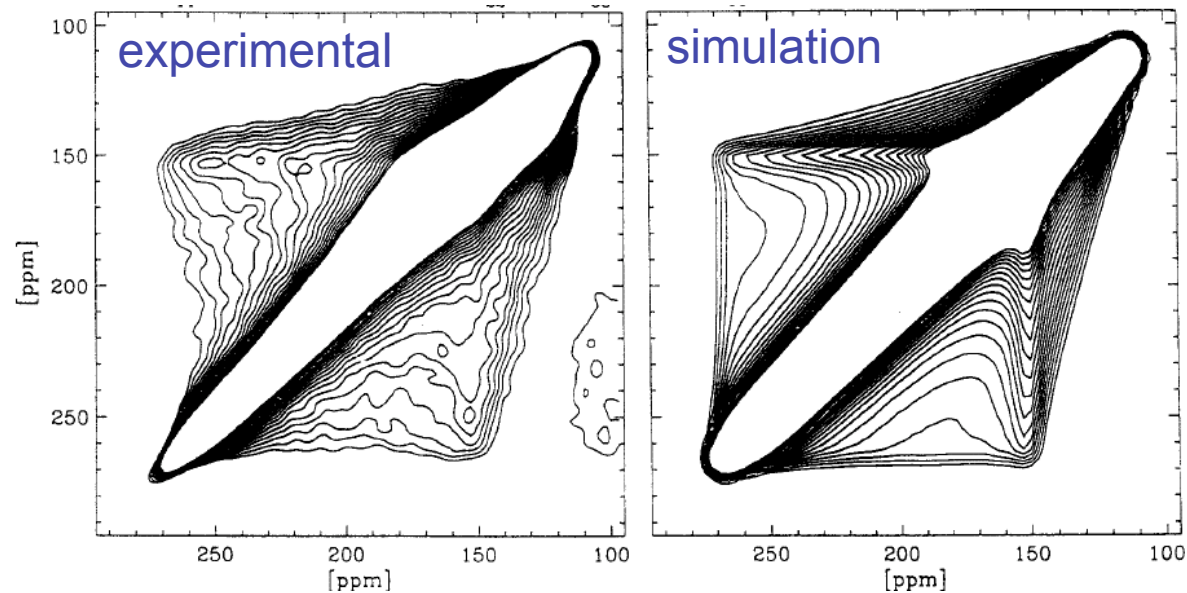


Exchange experiments



^{13}C NMR of PMMA

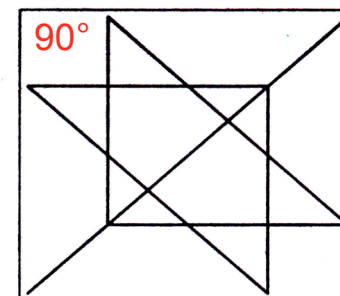
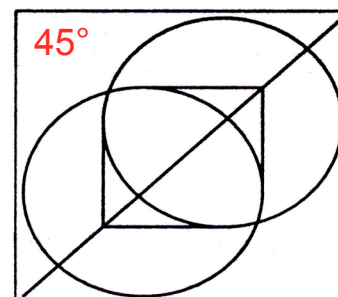
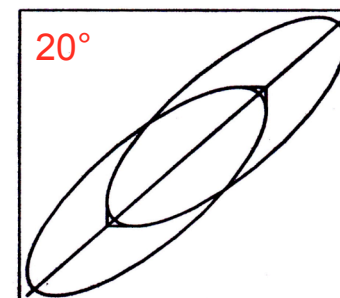
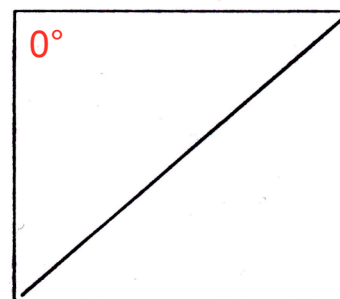
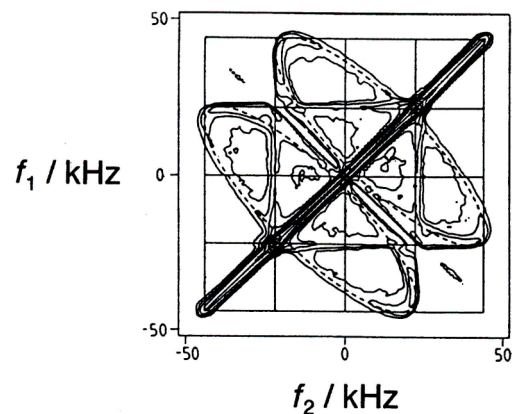
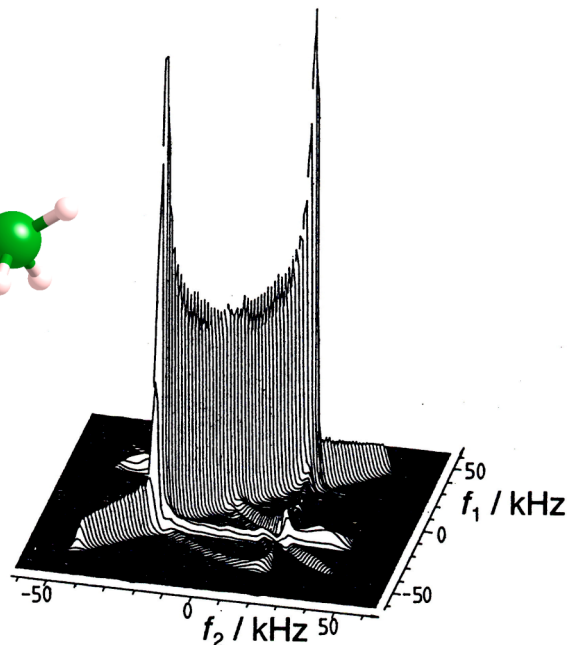
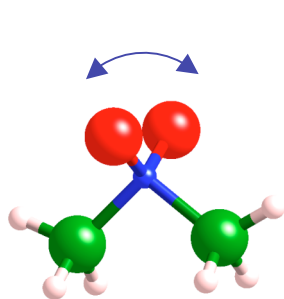
side chain flips of 180°



Schmidt Rohr et al.,
Macromol. **27**, 4733 (1994)

Exchange experiments

^2H exchange NMR of DMS



jumps around the C_2 axis of 106°

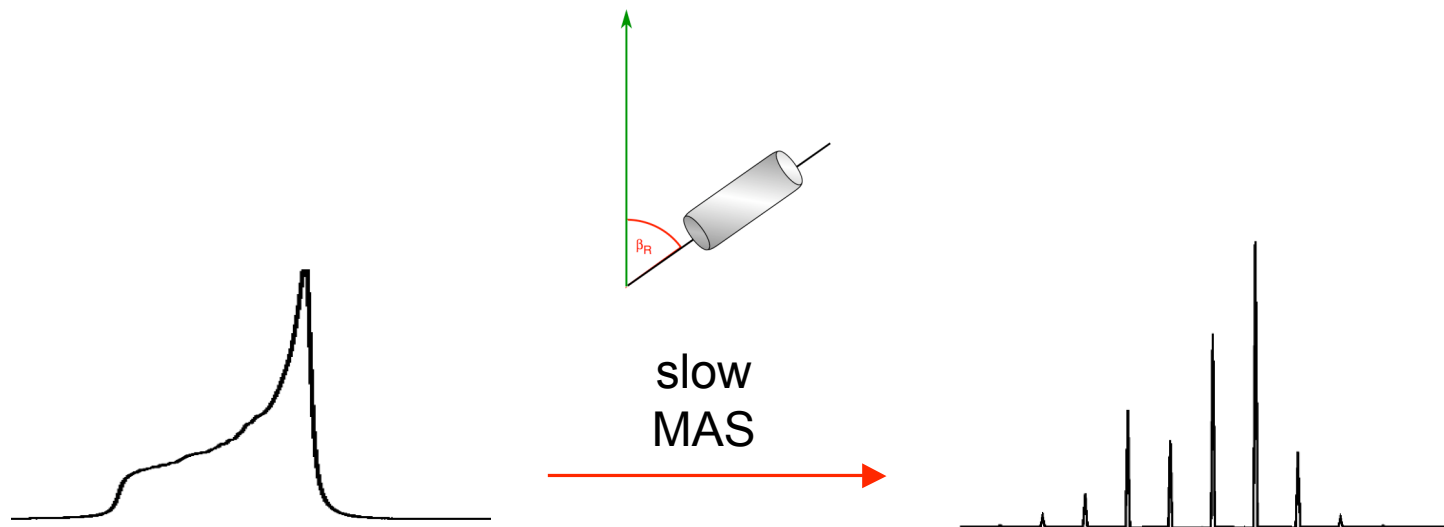
Slow MAS exchange experiments

Use MAS to improve resolution and sensitivity

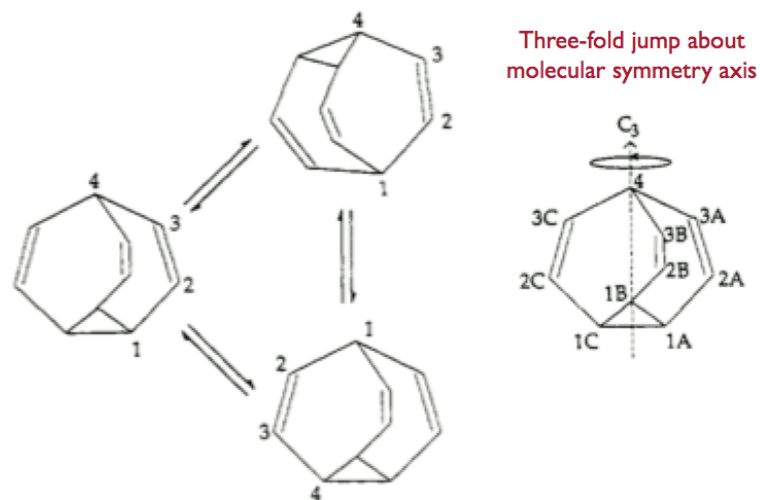
For slow MAS the interaction anisotropy is retained in spinning sidebands

Exchange peaks between different sites

Exchange peaks between sidebands of one site



Slow MAS exchange experiments

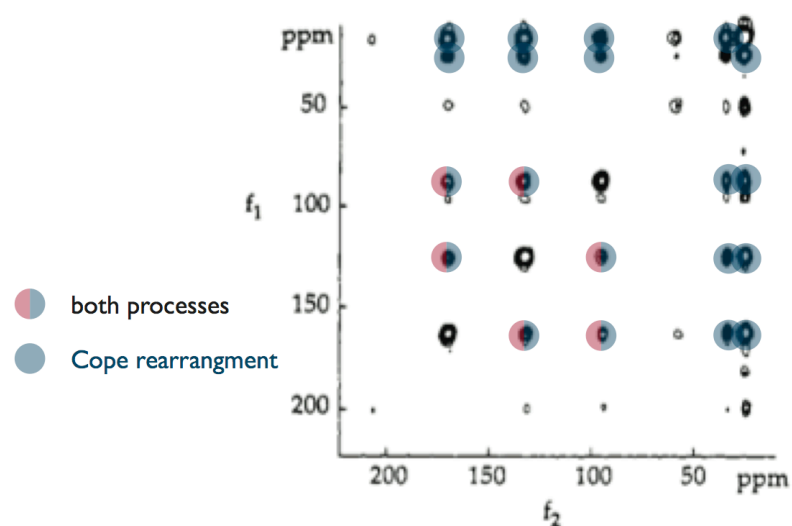
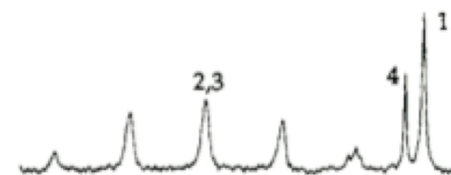


Cope rearrangement interconverts 10/3 tautomers coupled with a molecular reorientation

Two sets of cross peaks

1. Sidebands of different sites (Cope rearrangement, 15 kcal mol⁻¹)
2. Sidebands of same site (3 fold jumps, 21 kcal mol⁻¹)

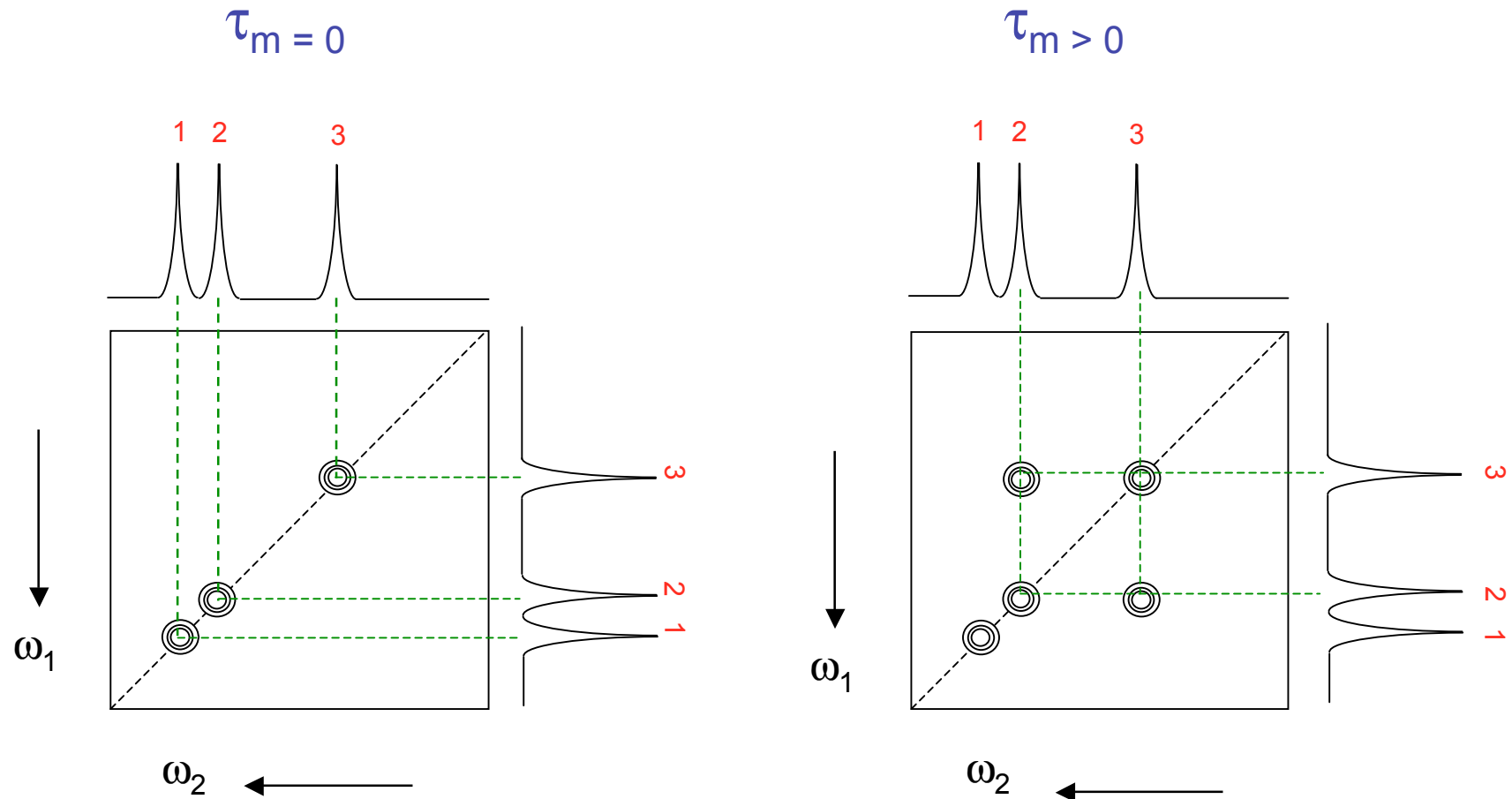
¹³C MAS NMR of bullvalene



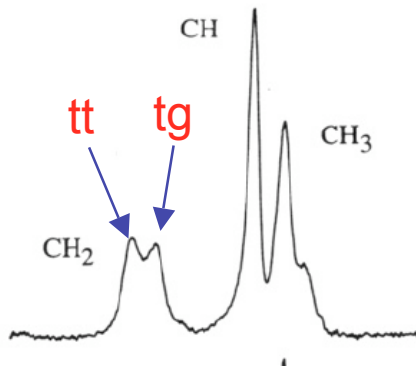
Titman et al. *J. Am. Chem. Soc.* **114**, 3765 (1992)

MAS exchange experiments

- Under fast MAS high resolution but no anisotropy information
- Exchange between different sites produces cross peaks



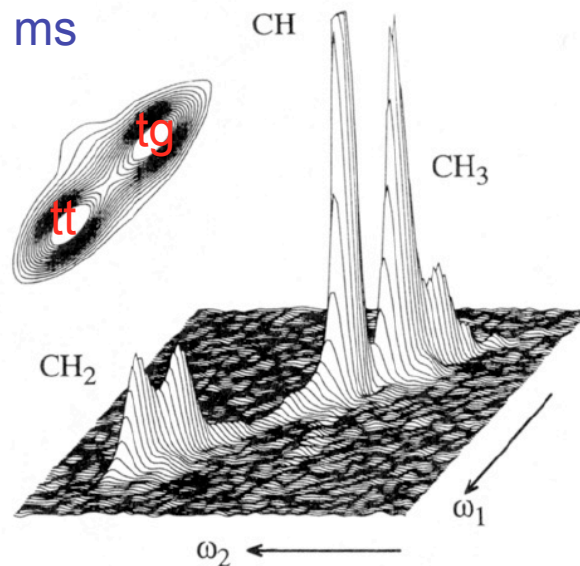
MAS exchange experiments



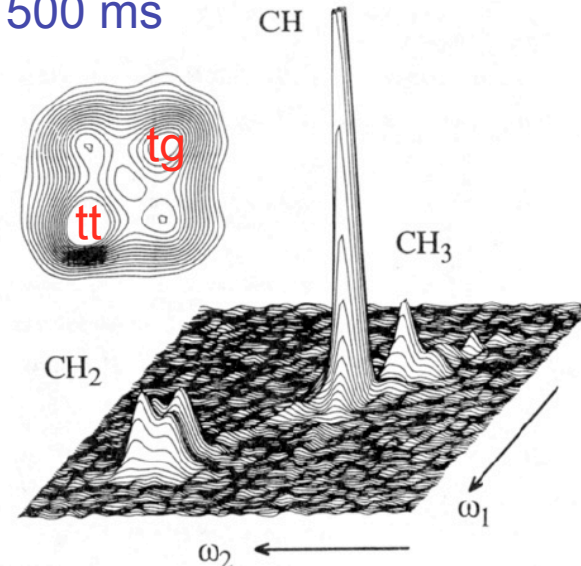
^{13}C MAS NMR atactic polypropylene

rotational motion of polymer backbone just below glass transition temperature interchanges different conformations

$\tau_m = 5 \text{ ms}$



$\tau_m = 500 \text{ ms}$



Timescales

